

# Diethylmalonic acid, isobutyl 2,3,6-trichlorophenyl ester

Inchi:	InChI=1S/C17H21Cl3O4/c1-5-17(6-2,15(21)23-9-10(3)4)16(22)24-14-12(19)8-7-11(18)13
InchiKey:	QZTXOSYZSUHOQZ-UHFFFAOYSA-N
Formula:	C17H21Cl3O4
SMILES:	CCC(CC)(C(=O)OCC(C)C)C(=O)Oc1c(Cl)ccc(Cl)c1Cl
Mol. weight [g/mol]:	395.70

## Physical Properties

Property code	Value	Unit	Source
gf	-327.45	kJ/mol	Joback Method
hf	-742.94	kJ/mol	Joback Method
hfus	39.89	kJ/mol	Joback Method
hvap	87.48	kJ/mol	Joback Method
log10ws	-5.99		Crippen Method
logp	5.558		Crippen Method
mcvol	278.230	ml/mol	McGowan Method
pc	1530.66	kPa	Joback Method
rinqol	2373.00		NIST Webbook
tb	891.18	K	Joback Method
tc	1114.69	K	Joback Method
tf	566.83	K	Joback Method
vc	1.058	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	778.49	J/molxK	891.18	Joback Method
cpg	828.03	J/molxK	1077.44	Joback Method
cpg	820.17	J/molxK	1040.19	Joback Method
cpg	811.32	J/molxK	1002.93	Joback Method
cpg	801.45	J/molxK	965.68	Joback Method
cpg	790.52	J/molxK	928.43	Joback Method
cpg	834.93	J/molxK	1114.69	Joback Method
dvisc	0.0000393	Paxs	891.18	Joback Method
dvisc	0.0000499	Paxs	837.12	Joback Method

dvisc	0.0000657	Paxs	783.06	Joback Method
dvisc	0.0000899	Paxs	729.00	Joback Method
dvisc	0.0001295	Paxs	674.95	Joback Method
dvisc	0.0001987	Paxs	620.89	Joback Method
dvisc	0.0003309	Paxs	566.83	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370283&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370283&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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