

# 2-Propenoic acid, oxiranylmethyl ester

<b>Other names:</b>	Acrylic acid, 2,3-epoxypropyl ester Glycidyl acrylate Glycidyl propenate 2,3-Epoxypropyl acrylate 2,3-Epoxypropanol acrylate M 581 Methacyclic acid, 2,3-epoxypropyl ester 1-Propanol, 2,3-epoxy-, acrylate Acrylic acid glycidyl ester Glycidylester kyseliny akrylove 2-Propenoic acid, 2-oxiranylmethyl ester NSC 24151
<b>Inchi:</b>	InChI=1S/C6H8O3/c1-2-6(7)9-4-5-3-8-5/h2,5H,1,3-4H2
<b>InchiKey:</b>	RPQRDASANLAFCM-UHFFFAOYSA-N
<b>Formula:</b>	C6H8O3
<b>SMILES:</b>	<chem>C=CC(=O)OCC1CO1</chem>
<b>Mol. weight [g/mol]:</b>	128.13
<b>CAS:</b>	106-90-1

## Physical Properties

Property code	Value	Unit	Source
gf	-171.81	kJ/mol	Joback Method
hf	-345.74	kJ/mol	Joback Method
hfus	18.92	kJ/mol	Joback Method
hvap	41.86	kJ/mol	Joback Method
log10ws	-0.14		Crippen Method
logp	0.114		Crippen Method
mcvol	93.550	ml/mol	McGowan Method
pc	3980.54	kPa	Joback Method
tb	443.34	K	Joback Method
tc	640.22	K	Joback Method
tf	272.29	K	Joback Method
vc	0.354	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	195.10	J/molxK	443.34	Joback Method
cpg	205.17	J/molxK	476.15	Joback Method
cpg	214.70	J/molxK	508.97	Joback Method
cpg	223.71	J/molxK	541.78	Joback Method
cpg	232.21	J/molxK	574.59	Joback Method
cpg	240.24	J/molxK	607.41	Joback Method
cpg	247.81	J/molxK	640.22	Joback Method
dvisc	0.0018399	Paxs	272.29	Joback Method
dvisc	0.0013500	Paxs	300.80	Joback Method
dvisc	0.0010450	Paxs	329.31	Joback Method
dvisc	0.0008427	Paxs	357.81	Joback Method
dvisc	0.0007014	Paxs	386.32	Joback Method
dvisc	0.0005988	Paxs	414.83	Joback Method
dvisc	0.0005217	Paxs	443.34	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	388.20	K	10.00	NIST Webbook

## Sources

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C106901&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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