

# Metamfepramone

<b>Other names:</b>	1-Propanone, 2-(dimethylamino)-1-phenyl- «alpha»-Dimethylaminopropiophenone Metamfepyramone MG 559 Propiophenone, 2-(dimethylamino)- 2-(Dimethylamino)propiofenone Dimethylpropion Metamfepyramon NSC 234706
<b>Inchi:</b>	InChI=1S/C11H15NO/c1-9(12(2)3)11(13)10-7-5-4-6-8-10/h4-9H,1-3H3
<b>InchiKey:</b>	KBHMHROOFHVLBA-UHFFFAOYSA-N
<b>Formula:</b>	C11H15NO
<b>SMILES:</b>	<chem>CC(C(=O)c1ccccc1)N(C)C</chem>
<b>Mol. weight [g/mol]:</b>	177.24
<b>CAS:</b>	15351-09-4

## Physical Properties

Property code	Value	Unit	Source
chl	-6337.70 ± 1.40	kJ/mol	NIST Webbook
gf	133.57	kJ/mol	Joback Method
hf	-69.90 ± 1.90	kJ/mol	NIST Webbook
hfl	-134.70 ± 1.50	kJ/mol	NIST Webbook
hfus	19.38	kJ/mol	Joback Method
hvap	64.80 ± 1.20	kJ/mol	NIST Webbook
hvap	64.80 ± 1.20	kJ/mol	NIST Webbook
log10ws	-2.07		Crippen Method
logp	1.819		Crippen Method
mcvol	153.640	ml/mol	McGowan Method
pc	2847.48	kPa	Joback Method
rinpol	1372.00		NIST Webbook
tb	543.63	K	Joback Method
tc	756.66	K	Joback Method
tf	307.55	K	Joback Method
vc	0.561	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	356.68	J/mol×K	543.63	Joback Method
cpg	372.53	J/mol×K	579.13	Joback Method
cpg	387.36	J/mol×K	614.64	Joback Method
cpg	401.22	J/mol×K	650.14	Joback Method
cpg	414.16	J/mol×K	685.65	Joback Method
cpg	426.23	J/mol×K	721.15	Joback Method
cpg	437.47	J/mol×K	756.66	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15351094&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15351094&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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