

# silver iodide

|                      |                              |
|----------------------|------------------------------|
| Inchi:               | InChI=1S/Ag.II/h;1H/q+1;/p-1 |
| InchiKey:            | MSFPLIAKTHOCQP-UHFFFAOYSA-M  |
| Formula:             | Agl                          |
| SMILES:              | [Ag+].[I-]                   |
| Mol. weight [g/mol]: | 234.77                       |
| CAS:                 | 7783-96-2                    |

## Physical Properties

| Property code | Value  | Unit | Source   |
|---------------|--------|------|--|
| ie            | 8.80   | eV   | NIST Webbook   |
| ie            | 8.90   | eV   | NIST Webbook   |
| ie            | 8.40   | eV   | NIST Webbook   |
| ie            | 13.75  | eV   | NIST Webbook   |
| ie            | 10.21  | eV   | NIST Webbook   |
| ie            | 8.80   | eV   | NIST Webbook   |
| ie            | 9.27   | eV   | NIST Webbook   |
| tf            | 832.00 | K    | Ultrasonic velocity for an equimolar mixture of molten Agl and NaCl in the biphasic region |
| tf            | 832.00 | K    | Densities of a dissolving mixture of molten (Agl + NaCl)                                   |
| tf            | 832.00 | K    | An anion effect on the separation of Agl-containing melts using sound waves                |
| tt            | 832.00 | K    | Phase-boundary potential in the two-liquid-phase (Agl + NaCl) system                       |

## Correlations

| Information   | Value                                 |
|---------------|---------------------------------------|
| Property code | pvap                                  |
| Equation      | $\ln(P_{\text{vap}}) = A + B/(T + C)$ |
| Coeff. A      | 1.39563e+01                           |

|                             |              |
|-----------------------------|--------------|
| Coeff. B                    | -1.54173e+04 |
| Coeff. C                    | -1.28120e+02 |
| Temperature range (K), min. | 1093.15      |
| Temperature range (K), max. | 1779.15      |

## Sources

**Densities of a dissolving mixture of molten (AgI + NaCl):**  
**An anion effect on the separation of AgI-containing melts using sound waves: boundary potential in the two-liquid-phase (AgI + NaCl) system:**  
**NIST Webbook:**

<https://www.doi.org/10.1016/j.jct.2013.03.027>

<https://www.doi.org/10.1016/j.jct.2015.03.022>

<https://www.doi.org/10.1016/j.jct.2015.06.015>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C7783962&Units=SI>

**The Yaws Handbook of Vapor Pressure:**  
**Ultrasonic velocity for an equimolar mixture of molten AgI and NaCl in the biphasic region:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<https://www.doi.org/10.1016/j.jct.2012.12.015>

## Legend

**ie:** Ionization energy

**pvap:** Vapor pressure

**tf:** Normal melting (fusion) point

**tt:** Triple Point Temperature

Latest version available from:

<https://www.chemeo.com/cid/23-899-9/silver-iodide.pdf>

Generated by Cheméo on 2024-05-18 20:22:27.002012046 +0000 UTC m=+18352995.922589368.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.