

# 2(3H)-Furanone, dihydro-5-propyl-

<b>Other names:</b>	1,4-Heptanolide 4-Heptanolide 4-Hydroxyheptanoic acid lactone 4-Hydroxyheptanoic acid, «gamma»-lactone 5-Propyldihydro-2(3H)-furanone Heptan-4-olide Heptanoic acid, 4-hydroxy-, «gamma»-lactone Heptanolide-4,1 NSC 46097 dihydro-5-propyl-2(3H)-furanone «gamma»-Heptalactone «gamma»-Heptanolactone «gamma»-Heptanolide «gamma»-Propiobutyrolactone «gamma»-Propyl-«gamma»-butyrolactone
<b>Inchi:</b>	InChI=1S/C7H12O2/c1-2-3-6-4-5-7(8)9-6/h6H,2-5H2,1H3
<b>InchiKey:</b>	VLSVVMPLPMNWBH-UHFFFAOYSA-N
<b>Formula:</b>	C7H12O2
<b>SMILES:</b>	CCCC1CCC(=O)O1
<b>Mol. weight [g/mol]:</b>	128.17
<b>CAS:</b>	105-21-5

## Physical Properties

Property code	Value	Unit	Source
gf	-164.10	kJ/mol	Joback Method
hf	-397.03	kJ/mol	Joback Method
hfus	15.31	kJ/mol	Joback Method
hvap	62.30 ± 0.30	kJ/mol	NIST Webbook
log10ws	-1.62		Crippen Method
logp	1.492		Crippen Method
mcvol	106.070	ml/mol	McGowan Method
pc	3526.27	kPa	Joback Method
rinpol	1128.00		NIST Webbook
rinpol	1127.00		NIST Webbook
rinpol	1159.00		NIST Webbook
rinpol	1103.00		NIST Webbook
rinpol	1172.00		NIST Webbook

rinpol	1113.00		NIST Webbook
rinpol	1103.00		NIST Webbook
rinpol	1155.00		NIST Webbook
rinpol	1103.00		NIST Webbook
rinpol	1101.00		NIST Webbook
rinpol	1163.00		NIST Webbook
rinpol	1157.00		NIST Webbook
rinpol	1163.00		NIST Webbook
ripol	1785.00		NIST Webbook
ripol	1764.00		NIST Webbook
ripol	1803.00		NIST Webbook
ripol	1785.00		NIST Webbook
ripol	1784.00		NIST Webbook
ripol	1797.00		NIST Webbook
ripol	1755.00		NIST Webbook
ripol	1811.00		NIST Webbook
ripol	1777.00		NIST Webbook
ripol	1784.00		NIST Webbook
ripol	1817.00		NIST Webbook
ripol	1796.00		NIST Webbook
ripol	1796.00		NIST Webbook
ripol	1789.00		NIST Webbook
ripol	1767.00		NIST Webbook
ripol	1803.00		NIST Webbook
ripol	1811.00		NIST Webbook
ripol	1766.00		NIST Webbook
tb	469.61	K	Joback Method
tc	683.50	K	Joback Method
tf	274.34	K	Joback Method
vc	0.397	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	234.28	J/mol×K	469.61	Joback Method
cpg	248.59	J/mol×K	505.26	Joback Method
cpg	262.27	J/mol×K	540.91	Joback Method
cpg	275.34	J/mol×K	576.56	Joback Method
cpg	287.80	J/mol×K	612.21	Joback Method
cpg	299.62	J/mol×K	647.86	Joback Method
cpg	310.83	J/mol×K	683.50	Joback Method

pvap	0.02	kPa	313.10	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.01	kPa	302.80	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.02	kPa	308.00	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	7.29e-03	kPa	298.10	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.03	kPa	318.00	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.05	kPa	323.10	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.07	kPa	328.10	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.09	kPa	333.00	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.13	kPa	338.20	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.18	kPa	343.20	Vapour pressures and enthalpies of vapourization of a series of the c-lactones

pvap	0.24	kPa	348.20	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.32	kPa	353.20	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.44	kPa	358.40	Vapour pressures and enthalpies of vapourization of a series of the c-lactones
pvap	0.57	kPa	363.20	Vapour pressures and enthalpies of vapourization of a series of the c-lactones

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Vapour pressures and enthalpies of vapourization of a series of the c-lactones:</b>	<a href="https://www.doi.org/10.1016/j.jct.2008.02.002">https://www.doi.org/10.1016/j.jct.2008.02.002</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C105215&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C105215&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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