

# 2-(Diethylamino)acetonitrile

<b>Other names:</b>	Diethylaminoacetonitrile Acetonitrile, (diethylamino)- Glycinonitrile, N,N-diethyl- N,N-Diethylaminoacetonitrile Nitril kiseliny diethylaminooctove
<b>Inchi:</b>	InChI=1S/C6H12N2/c1-3-8(4-2)6-5-7/h3-4,6H2,1-2H3
<b>InchiKey:</b>	LVPZSMIBSMMLPI-UHFFFAOYSA-N
<b>Formula:</b>	C6H12N2
<b>SMILES:</b>	CCN(CC)CC#N
<b>Mol. weight [g/mol]:</b>	112.17
<b>CAS:</b>	3010-02-4

## Physical Properties

Property code	Value	Unit	Source
chl	-4092.15 ± 0.85	kJ/mol	NIST Webbook
gf	243.60	kJ/mol	Joback Method
hf	66.00 ± 1.00	kJ/mol	NIST Webbook
hfl	16.10 ± 1.00	kJ/mol	NIST Webbook
hfus	15.82	kJ/mol	Joback Method
hvap	49.93 ± 0.27	kJ/mol	NIST Webbook
hvap	49.90	kJ/mol	NIST Webbook
log10ws	-0.77		Crippen Method
logp	0.852		Crippen Method
mcvol	106.760	ml/mol	McGowan Method
pc	3015.64	kPa	Joback Method
rinpol	912.00		NIST Webbook
tb	443.20	K	NIST Webbook
tc	638.03	K	Joback Method
tf	254.84	K	Joback Method
vc	0.415	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	266.53	J/mol×K	606.89	Joback Method
cpg	219.32	J/mol×K	451.20	Joback Method
cpg	229.70	J/mol×K	482.34	Joback Method
cpg	239.60	J/mol×K	513.48	Joback Method
cpg	249.03	J/mol×K	544.61	Joback Method
cpg	258.00	J/mol×K	575.75	Joback Method
cpg	274.63	J/mol×K	638.03	Joback Method
hvapt	49.90 ± 0.30	kJ/mol	300.50	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	343.20	K	3.20	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3010024&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3010024&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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