

# Carbonic acid, propyl 4-benzyloxyphenyl ester

<b>Inchi:</b>	InChI=1S/C17H18O4/c1-2-12-19-17(18)21-16-10-8-15(9-11-16)20-13-14-6-4-3-5-7-14/h3
<b>InchiKey:</b>	UZEZQAFDXMYNHE-UHFFFAOYSA-N
<b>Formula:</b>	C17H18O4
<b>SMILES:</b>	CCCOC(=O)Oc1ccc(OCc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	286.32

## Physical Properties

Property code	Value	Unit	Source
gf	-136.47	kJ/mol	Joback Method
hf	-441.86	kJ/mol	Joback Method
hfus	32.64	kJ/mol	Joback Method
hvap	72.63	kJ/mol	Joback Method
log10ws	-4.92		Crippen Method
logp	4.191		Crippen Method
mvol	222.050	ml/mol	McGowan Method
pc	2083.12	kPa	Joback Method
rinpol	2312.00		NIST Webbook
rinpol	2312.00		NIST Webbook
tb	767.83	K	Joback Method
tc	991.87	K	Joback Method
tf	463.33	K	Joback Method
vc	0.832	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	625.53	J/molxK	767.83	Joback Method
cpg	640.78	J/molxK	805.17	Joback Method
cpg	654.81	J/molxK	842.51	Joback Method
cpg	667.62	J/molxK	879.85	Joback Method
cpg	679.24	J/molxK	917.19	Joback Method
cpg	689.67	J/molxK	954.53	Joback Method
cpg	698.94	J/molxK	991.87	Joback Method
dvisc	0.0005696	Paxs	463.33	Joback Method

dvisc	0.0003316	Paxs	514.08	Joback Method
dvisc	0.0002128	Paxs	564.83	Joback Method
dvisc	0.0001469	Paxs	615.58	Joback Method
dvisc	0.0001073	Paxs	666.33	Joback Method
dvisc	0.0000819	Paxs	717.08	Joback Method
dvisc	0.0000648	Paxs	767.83	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357824&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357824&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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