

Benzaldehyde, 4-(dimethylamino)-

Other names:	Benzaldehyde, p-(dimethylamino)- p-(Dimethylamino)benzaldehyde p-(N,N-Dimethylamino)benzaldehyde p-Formyl-N,N-dimethylaniline p-Formyldimethylaniline Ehrlich's reagent 4-(Dimethylamino)benzaldehyde 4-Dimethylaminobenzenecarbonal para-Dimethylaminobenzaldehyde 4-N,N-Dimethylaminobenzaldehyde N,N-Dimethyl-p-aminobenzaldehyde Reagens ehrlichovo 4-Formyl-N,N-dimethylaniline N,N-Dimethyl-4-formylaniline N,N-Dimethyl-4-aminobenzaldehyde p-DAB NSC 5517
Inchi:	InChI=1S/C9H11NO/c1-10(2)9-5-3-8(7-11)4-6-9/h3-7H,1-2H3
InchiKey:	BGNGWHSBYQYVRX-UHFFFAOYSA-N
Formula:	C9H11NO
SMILES:	CN(C)c1ccc(C=O)cc1
Mol. weight [g/mol]:	149.19
CAS:	100-10-7

Physical Properties

Property code	Value	Unit	Source
affp	924.80	kJ/mol	NIST Webbook
basg	898.30	kJ/mol	NIST Webbook
chs	-4977.00	kJ/mol	NIST Webbook
chs	-4976.03 ± 0.84	kJ/mol	NIST Webbook
gf	138.94	kJ/mol	Joback Method
hf	-22.08	kJ/mol	Joback Method
hfs	-137.00	kJ/mol	NIST Webbook
hfus	18.03	kJ/mol	Joback Method
hvap	47.33	kJ/mol	Joback Method
ie	7.30 ± 0.10	eV	NIST Webbook
ie	7.81	eV	NIST Webbook

ie	7.36 ± 0.02	eV	NIST Webbook
ie	7.30 ± 0.10	eV	NIST Webbook
log10ws	-1.62		Crippen Method
logp	1.565		Crippen Method
mcvol	125.460	ml/mol	McGowan Method
pc	3484.76	kPa	Joback Method
rinpol	1528.00		NIST Webbook
rinpol	1481.00		NIST Webbook
rinpol	1528.00		NIST Webbook
rinpol	1528.00		NIST Webbook
rinpol	1481.00		NIST Webbook
rinpol	1528.00		NIST Webbook
ripol	2459.00		NIST Webbook
ripol	2459.00		NIST Webbook
tb	498.08	K	Joback Method
tc	709.02	K	Joback Method
tf	347.00	K	NIST Webbook
vc	0.467	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	324.73	J/mol×K	673.86	Joback Method
cpg	267.08	J/mol×K	498.08	Joback Method
cpg	280.15	J/mol×K	533.24	Joback Method
cpg	292.41	J/mol×K	568.39	Joback Method
cpg	303.91	J/mol×K	603.55	Joback Method
cpg	314.67	J/mol×K	638.70	Joback Method
cpg	334.13	J/mol×K	709.02	Joback Method
hfust	19.07	kJ/mol	346.20	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	449.70	K	2.30	NIST Webbook
tbrp	418.00	K	1.30	NIST Webbook

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C100107&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

affp:	Proton affinity
basg:	Gas basicity
chs:	Standard solid enthalpy of combustion
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfs:	Solid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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