

# 2-Fluorobenzoic acid, 3,4-dichlorophenyl ester

<b>Inchi:</b>	InChI=1S/C13H7Cl2FO2/c14-10-6-5-8(7-11(10)15)18-13(17)9-3-1-2-4-12(9)16/h1-7H
<b>InchiKey:</b>	ODWCIDDVKYMMDX-UHFFFAOYSA-N
<b>Formula:</b>	C13H7Cl2FO2
<b>SMILES:</b>	O=C(Oc1ccc(Cl)c(Cl)c1)c1ccccc1F
<b>Mol. weight [g/mol]:</b>	285.10

## Physical Properties

Property code	Value	Unit	Source
gf	-198.08	kJ/mol	Joback Method
hf	-345.39	kJ/mol	Joback Method
hfus	30.60	kJ/mol	Joback Method
hvap	68.18	kJ/mol	Joback Method
log10ws	-5.26		Crippen Method
logp	4.352		Crippen Method
mcvol	180.200	ml/mol	McGowan Method
pc	2707.03	kPa	Joback Method
rinpola	2012.00		NIST Webbook
rinpola	2012.00		NIST Webbook
tb	715.56	K	Joback Method
tc	959.61	K	Joback Method
tf	459.26	K	Joback Method
vc	0.688	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	416.98	J/mol×K	715.56	Joback Method
cpg	428.00	J/mol×K	756.24	Joback Method
cpg	438.06	J/mol×K	796.91	Joback Method
cpg	447.19	J/mol×K	837.59	Joback Method
cpg	455.42	J/mol×K	878.26	Joback Method
cpg	462.77	J/mol×K	918.94	Joback Method
cpg	469.29	J/mol×K	959.61	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307690&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307690&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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