

# 2-Aminobenzyl alcohol

<b>Other names:</b>	Benzyl alcohol, o-amino- o-(Hydroxymethyl)aniline o-Aminobenzyl alcohol o-Aminobenzyl alcohol o-Aminobenzyl alcohol o-Aminophenylcarbinol Benzenemethanol, 2-amino- 2-Aminobenzenemethanol (2-Aminophenyl)methanol NSC 1173 (o-Aminophenyl)methanol
<b>Inchi:</b>	InChI=1S/C7H9NO/c8-7-4-2-1-3-6(7)5-9/h1-4,9H,5,8H2
<b>InchiKey:</b>	VYFOAVADNIHPTR-UHFFFAOYSA-N
<b>Formula:</b>	C7H9NO
<b>SMILES:</b>	<chem>Nc1ccccc1CO</chem>
<b>Mol. weight [g/mol]:</b>	123.15
<b>CAS:</b>	5344-90-1

## Physical Properties

Property code	Value	Unit	Source
gf	40.47	kJ/mol	Joback Method
hf	-81.19	kJ/mol	Joback Method
hfus	16.82	kJ/mol	Joback Method
hvap	61.43	kJ/mol	Joback Method
log10ws	-1.25		Crippen Method
logp	0.761		Crippen Method
mcvol	101.580	ml/mol	McGowan Method
pc	4924.59	kPa	Joback Method
tb	548.20	K	NIST Webbook
tb	548.00 ± 5.00	K	NIST Webbook
tc	767.32	K	Joback Method
tf	357.00	K	NIST Webbook
vc	0.367	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	230.88	J/mol×K	555.93	Joback Method
cpg	240.23	J/mol×K	591.16	Joback Method
cpg	249.02	J/mol×K	626.39	Joback Method
cpg	257.26	J/mol×K	661.62	Joback Method
cpg	264.99	J/mol×K	696.85	Joback Method
cpg	272.21	J/mol×K	732.09	Joback Method
cpg	278.97	J/mol×K	767.32	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	435.20	K	2.00	NIST Webbook
tbrp	433.00	K	1.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5344901&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5344901&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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