

# 2,2-Bis(p-chlorophenyl)ethanol

**Other names:**

Benzeneethanol, 4-chloro-«beta»-(4-chlorophenyl)-  
Ethanol, 2,2-bis(p-chlorophenyl)-  
DDOH  
DDOM  
2,2-Bis(4-chlorophenyl)-1-hydroxyethane  
2,2-Bis(4-chlorophenyl)ethanol  
p,p'-DDOH  
Ethanol, 2,2-bis-(4-chlorophenyl)  
NSC 8942

**Inchi:**

InChI=1S/C14H12Cl2O/c15-12-5-1-10(2-6-12)14(9-17)11-3-7-13(16)8-4-11/h1-8,14,17H,

**InchiKey:**

ZVIDYKRNLNAXFT-UHFFFAOYSA-N

**Formula:**

C14H12Cl2O

**SMILES:**

OCC(c1ccc(Cl)cc1)c1ccc(Cl)cc1

**Mol. weight [g/mol]:**

267.15

**CAS:**

2642-82-2

## Physical Properties

Property code	Value	Unit	Source
gf	109.44	kJ/mol	Joback Method
hf	-71.16	kJ/mol	Joback Method
hfus	28.28	kJ/mol	Joback Method
hvap	77.69	kJ/mol	Joback Method
log10ws	-4.59		Crippen Method
logp	4.118		Crippen Method
mcvol	190.950	ml/mol	McGowan Method
pc	2758.46	kPa	Joback Method
rinpol	2186.00		NIST Webbook
rinpol	2186.00		NIST Webbook
tb	749.64	K	Joback Method
tc	980.91	K	Joback Method
tf	431.08	K	Joback Method
vc	0.715	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	474.14	J/molxK	749.64	Joback Method
cpg	523.12	J/molxK	942.37	Joback Method
cpg	514.89	J/molxK	903.82	Joback Method
cpg	505.94	J/molxK	865.28	Joback Method
cpg	496.21	J/molxK	826.73	Joback Method
cpg	485.63	J/molxK	788.19	Joback Method
cpg	530.69	J/molxK	980.91	Joback Method
dvisc	0.0000319	Paxs	749.64	Joback Method
dvisc	0.0000464	Paxs	696.55	Joback Method
dvisc	0.0000719	Paxs	643.45	Joback Method
dvisc	0.0001205	Paxs	590.36	Joback Method
dvisc	0.0002237	Paxs	537.27	Joback Method
dvisc	0.0004753	Paxs	484.17	Joback Method
dvisc	0.0012160	Paxs	431.08	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2642822&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2642822&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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