

# Cyclopropanecarboxylic acid, 3-ethylphenyl ester

<b>Inchi:</b>	InChI=1S/C12H14O2/c1-2-9-4-3-5-11(8-9)14-12(13)10-6-7-10/h3-5,8,10H,2,6-7H2,1H3
<b>InchiKey:</b>	GQXFHKNHQKEMW-UHFFFAOYSA-N
<b>Formula:</b>	C12H14O2
<b>SMILES:</b>	CCc1cccc(OC(=O)C2CC2)c1
<b>Mol. weight [g/mol]:</b>	190.24

## Physical Properties

Property code	Value	Unit	Source
gf	-20.23	kJ/mol	Joback Method
hf	-237.95	kJ/mol	Joback Method
hfus	21.41	kJ/mol	Joback Method
hvap	54.31	kJ/mol	Joback Method
log10ws	-3.08		Crippen Method
logp	2.564		Crippen Method
mvol	152.760	ml/mol	McGowan Method
pc	2835.36	kPa	Joback Method
rinpol	1538.00		NIST Webbook
rinpol	1538.00		NIST Webbook
tb	588.65	K	Joback Method
tc	810.65	K	Joback Method
tf	354.04	K	Joback Method
vc	0.581	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	379.28	J/mol×K	588.65	Joback Method
cpg	394.83	J/mol×K	625.65	Joback Method
cpg	409.38	J/mol×K	662.65	Joback Method
cpg	422.97	J/mol×K	699.65	Joback Method
cpg	435.64	J/mol×K	736.65	Joback Method
cpg	447.46	J/mol×K	773.65	Joback Method
cpg	458.48	J/mol×K	810.65	Joback Method
dvisc	0.0017722	Paxs	354.04	Joback Method

dvisc	0.0012387	Paxs	393.14	Joback Method
dvisc	0.0009238	Paxs	432.24	Joback Method
dvisc	0.0007232	Paxs	471.35	Joback Method
dvisc	0.0005879	Paxs	510.45	Joback Method
dvisc	0.0004922	Paxs	549.55	Joback Method
dvisc	0.0004219	Paxs	588.65	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U354661&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U354661&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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