

# Benzenamine, N-methyl-2,4-dinitro-

<b>Other names:</b>	Aniline, N-methyl-2,4-dinitro- N-Methyl-2,4-dinitroaniline 2,4-Dinitro-N-methylaniline (2,4-Dinitrophenyl)methylamine
<b>Inchi:</b>	InChI=1S/C7H7N3O4/c1-8-6-3-2-5(9(11)12)4-7(6)10(13)14/h2-4,8H,1H3
<b>InchiKey:</b>	IQEJEZOCXWJNKR-UHFFFAOYSA-N
<b>Formula:</b>	C7H7N3O4
<b>SMILES:</b>	CNc1ccc([N+](=O)[O-])cc1[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	197.15
<b>CAS:</b>	2044-88-4

## Physical Properties

Property code	Value	Unit	Source
chs	-3705.00	kJ/mol	NIST Webbook
chs	-3688.00 ± 3.70	kJ/mol	NIST Webbook
gf	261.70	kJ/mol	Joback Method
hf	57.73	kJ/mol	Joback Method
hfs	-67.40 ± 3.70	kJ/mol	NIST Webbook
hfus	34.97	kJ/mol	Joback Method
hvap	74.39	kJ/mol	Joback Method
log10ws	-2.78		Crippen Method
logp	1.545		Crippen Method
mcvol	130.550	ml/mol	McGowan Method
pc	4151.61	kPa	Joback Method
tb	750.05	K	Joback Method
tc	1018.88	K	Joback Method
tf	559.99	K	Joback Method
vc	0.518	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	343.20	J/mol×K	750.05	Joback Method
cpg	352.67	J/mol×K	794.86	Joback Method

cpg	361.22	J/mol×K	839.66	Joback Method
cpg	368.92	J/mol×K	884.47	Joback Method
cpg	375.82	J/mol×K	929.27	Joback Method
cpg	381.97	J/mol×K	974.08	Joback Method
cpg	387.43	J/mol×K	1018.88	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2044884&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2044884&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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