

# 2-Octanol, acetate

<b>Other names:</b>	1-Methylheptyl acetate 2-Acetoxyoctane 2-Octyl acetate Acetic acid, oct-2-yl ester Capryl acetate NSC 65620 Sec-octyl acetate
<b>Inchi:</b>	InChI=1S/C10H20O2/c1-4-5-6-7-8-9(2)12-10(3)11/h9H,4-8H2,1-3H3
<b>InchiKey:</b>	SJFUDWKNZGXSLV-UHFFFAOYSA-N
<b>Formula:</b>	C10H20O2
<b>SMILES:</b>	CCCCCCC(C)OC(C)=O
<b>Mol. weight [g/mol]:</b>	172.26
<b>CAS:</b>	2051-50-5

## Physical Properties

Property code	Value	Unit	Source
gf	-203.04	kJ/mol	Joback Method
hf	-499.81	kJ/mol	Joback Method
hfus	20.92	kJ/mol	Joback Method
hvap	46.62	kJ/mol	Joback Method
log10ws	-2.98		Crippen Method
logp	2.908		Crippen Method
mvol	159.200	ml/mol	McGowan Method
pc	2216.62	kPa	Joback Method
rinpol	1147.00		NIST Webbook
rinpol	1121.00		NIST Webbook
rinpol	1147.00		NIST Webbook
rinpol	1127.00		NIST Webbook
rinpol	1117.00		NIST Webbook
rinpol	1082.00		NIST Webbook
rinpol	1132.00		NIST Webbook
rinpol	1121.00		NIST Webbook
ripol	1497.00		NIST Webbook
ripol	1493.00		NIST Webbook
ripol	1493.00		NIST Webbook
tb	504.05	K	Joback Method
tc	679.81	K	Joback Method

tf	259.62	K	Joback Method
vc	0.614	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	368.31	J/mol×K	504.05	Joback Method
cpg	382.82	J/mol×K	533.34	Joback Method
cpg	396.77	J/mol×K	562.64	Joback Method
cpg	410.17	J/mol×K	591.93	Joback Method
cpg	423.03	J/mol×K	621.22	Joback Method
cpg	435.35	J/mol×K	650.52	Joback Method
cpg	447.14	J/mol×K	679.81	Joback Method
dvisc	0.0046674	Paxs	259.62	Joback Method
dvisc	0.0019627	Paxs	300.36	Joback Method
dvisc	0.0010151	Paxs	341.10	Joback Method
dvisc	0.0006043	Paxs	381.84	Joback Method
dvisc	0.0003976	Paxs	422.57	Joback Method
dvisc	0.0002816	Paxs	463.31	Joback Method
dvisc	0.0002109	Paxs	504.05	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47357e+01
Coeff. B	-4.09206e+03
Coeff. C	-7.12960e+01
Temperature range (K), min.	354.52
Temperature range (K), max.	505.50

## Sources

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="https://webbook.nist.gov/cgi/cbook.cgi?ID=C2051505&amp;Units=SI">https://webbook.nist.gov/cgi/cbook.cgi?ID=C2051505&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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