

# nickel dibromide

Other names:	nickel bromide nickel(II) bromide
Inchi:	InChI=1S/2BrH.Ni/h2*1H;/q;;+2/p-2
InchiKey:	IPLJNQFXJUCRNH-UHFFFAOYSA-L
Formula:	Br2Ni
SMILES:	[Br-].[Br-].[Ni+2]
Mol. weight [g/mol]:	218.50
CAS:	13462-88-9

## Physical Properties

Property code	Value	Unit	Source
hsub	226.00 ± 1.00	kJ/mol	NIST Webbook
ie	10.30	eV	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hsubt	207.00 ± 4.00	kJ/mol	841.50	NIST Webbook

## Sources

Thermochemistry of adducts of some bivalent transition metal bromides with triphenylphosphine.	<a href="https://www.doi.org/10.1016/j.tca.2006.05.022">https://www.doi.org/10.1016/j.tca.2006.05.022</a>
Thermochemistry of adducts of some bivalent transition metal bromides with triphenylphosphine.	<a href="https://www.doi.org/10.1016/j.tca.2007.01.034">https://www.doi.org/10.1016/j.tca.2007.01.034</a>
Thermochemistry of adducts of some bivalent transition metal bromides with triphenylphosphine.	<a href="https://www.doi.org/10.1016/j.tca.2007.11.018">https://www.doi.org/10.1016/j.tca.2007.11.018</a>
Thermodynamic Properties of Inorganic Salts in Nonaqueous Solutions. V. Apparent Molar Volumes and Compressibilities of Divalent Transition Metal Bromides in N,N-Dimethylformamide.	<a href="https://www.doi.org/10.1021/je8001877">https://www.doi.org/10.1021/je8001877</a>
NIST Webbook	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13462889&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13462889&amp;Units=SI</a>
Thermochemistry of adducts of some bivalent transition metal bromides with triphenylphosphine.	<a href="https://www.doi.org/10.1016/j.tca.2005.06.016">https://www.doi.org/10.1016/j.tca.2005.06.016</a>

# Legend

<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>ie:</b>	Ionization energy

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