

# Pyridine-3,5-dicarboxylic acid, 1,4-dihydro-2,6-dimethyl-4-(o-trifluoromethylphenyl) diethyl ester

InChI: InChI=1S/C20H22F3NO4/c1-5-27-18(25)15-11(3)24-12(4)16(19(26)28-6-2)17(15)13-9-7  
InChIKey: ACZJZGLXRPZSLI-UHFFFAOYSA-N

**Formula:** C20H22F3NO4  
**SMILES:** CCOC(=O)C1=C(C)NC(C)=C(C(=O)OCC)C1c1ccccc1C(F)(F)F  
**Mol. weight [g/mol]:** 397.39  
**CAS:** 23191-75-5

## Physical Properties

Property code	Value	Unit	Source
gf	-695.57	kJ/mol	Joback Method
hf	-1155.94	kJ/mol	Joback Method
hfus	50.92	kJ/mol	Joback Method
hvap	88.04	kJ/mol	Joback Method
log10ws	-5.45		Crippen Method
logp	4.066		Crippen Method
mcvol	279.610	ml/mol	McGowan Method
pc	1474.75	kPa	Joback Method
tb	922.16	K	Joback Method
tc	1141.57	K	Joback Method
tf	666.62	K	Joback Method
vc	1.081	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	883.13	J/molxK	922.16	Joback Method
cpg	895.96	J/molxK	958.73	Joback Method
cpg	907.45	J/molxK	995.30	Joback Method
cpg	917.61	J/molxK	1031.86	Joback Method
cpg	926.47	J/molxK	1068.43	Joback Method
cpg	934.04	J/molxK	1105.00	Joback Method
cpg	940.35	J/molxK	1141.57	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C23191755&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C23191755&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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