

# Benzenamine, 4-bromo-N,N-dimethyl-

<b>Other names:</b>	Aniline, p-bromo-N,N-dimethyl- p-Bromo(dimethylamino)benzene 1-Bromo-4-(dimethylamino)benzene p-Bromo-N,N-dimethylaniline 4-Bromo-N,N-dimethylaniline 4-Bromo-N,N-dimethylbenzenamine p-Dimethylaminobromobenzene p-N,N-Dimethylaminobromobenzene 4-Dimethylaminobromobenzene p-(Dimethylamino)phenyl bromide 4-(Dimethylamino)phenyl bromide N,N-Dimethyl-p-bromoaniline N,N-Dimethyl-4-bromoaniline Aniline, 4-bromo-N,N-dimethyl- (4-Bromophenyl)dimethylamine NSC 8056
<b>Inchi:</b>	InChI=1S/C8H10BrN/c1-10(2)8-5-3-7(9)4-6-8/h3-6H,1-2H3
<b>InchiKey:</b>	XYZWVMVYYUIMRIZ-UHFFFAOYSA-N
<b>Formula:</b>	C8H10BrN
<b>SMILES:</b>	CN(C)c1ccc(Br)cc1
<b>Mol. weight [g/mol]:</b>	200.08
<b>CAS:</b>	586-77-6

## Physical Properties

Property code	Value	Unit	Source
gf	244.36	kJ/mol	Joback Method
hf	110.47	kJ/mol	Joback Method
hfus	18.43	kJ/mol	Joback Method
hvap	44.82	kJ/mol	Joback Method
ie	7.33	eV	NIST Webbook
log10ws	-2.54		Crippen Method
logp	2.515		Crippen Method
mcvol	127.300	ml/mol	McGowan Method
pc	3891.64	kPa	Joback Method
tb	537.20	K	NIST Webbook
tc	719.00	K	Joback Method
tf	311.13	K	Joback Method

vc

0.456

m<sup>3</sup>/kmol

Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	244.37	J/mol×K	492.70	Joback Method
cpg	257.10	J/mol×K	530.42	Joback Method
cpg	268.95	J/mol×K	568.13	Joback Method
cpg	279.97	J/mol×K	605.85	Joback Method
cpg	290.20	J/mol×K	643.57	Joback Method
cpg	299.69	J/mol×K	681.29	Joback Method
cpg	308.50	J/mol×K	719.00	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C586776&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C586776&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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