

# Cyclopropane-1-carboxylic acid, 2-methoxymethyl-3-phenyl, ethyl ester

Inchi:	InChI=1S/C14H18O3/c1-3-17-14(15)13-11(9-16-2)12(13)10-7-5-4-6-8-10/h4-8,11-13H,3,
InchiKey:	XTXDOAAMPZLWCV-UHFFFAOYSA-N
Formula:	C14H18O3
SMILES:	CCOC(=O)C1C(COC)C1c1ccccc1
Mol. weight [g/mol]:	234.29

## Physical Properties

Property code	Value	Unit	Source
gf	-114.18	kJ/mol	Joback Method
hf	-440.66	kJ/mol	Joback Method
hfus	30.31	kJ/mol	Joback Method
hvap	59.89	kJ/mol	Joback Method
log10ws	-2.11		Crippen Method
logp	2.226		Crippen Method
mcvol	186.810	ml/mol	McGowan Method
pc	2187.68	kPa	Joback Method
rinpol	1521.00		NIST Webbook
tb	642.51	K	Joback Method
tc	853.50	K	Joback Method
tf	377.81	K	Joback Method
vc	0.709	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	510.90	J/molxK	642.51	Joback Method
cpg	587.56	J/molxK	818.33	Joback Method
cpg	574.21	J/molxK	783.17	Joback Method
cpg	559.90	J/molxK	748.00	Joback Method
cpg	544.60	J/molxK	712.84	Joback Method
cpg	528.27	J/molxK	677.67	Joback Method
cpg	599.96	J/molxK	853.50	Joback Method
dvisc	0.0004213	Paxs	642.51	Joback Method
dvisc	0.0004802	Paxs	598.39	Joback Method

dvisc	0.0005590	Paxs	554.28	Joback Method
dvisc	0.0006679	Paxs	510.16	Joback Method
dvisc	0.0008254	Paxs	466.04	Joback Method
dvisc	0.0010663	Paxs	421.93	Joback Method
dvisc	0.0014623	Paxs	377.81	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R329496&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R329496&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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