Cyclopropane, 1,1-dimethyl-2-(1-methylethenyl)-

Inchi: InChI=1S/C8H14/c1-6(2)7-5-8(7,3)4/h7H,1,5H2,2-4H3
InchiKey: LWEPLBXTZQHKEX-UHFFFAOYSA-N
Formula: C8H14
SMILES: C=C(C)C1CC1(C)C
Mol. weight [g/mol]: 110.20
CAS: 1121-35-3

Physical Properties

Property code	Value	Unit	Source
gf	143.32	kJ/mol	Joback Method
hf	-25.11	kJ/mol	Joback Method
hfus	6.79	kJ/mol	Joback Method
hvap	31.27	kJ/mol	Joback Method
ie	8.46	eV	NIST Webbook
log10ws	-2.44		Crippen Method
logp	2.609		Crippen Method
mcvol	108.420	ml/mol	McGowan Method
pc	3096.73	kPa	Joback Method
tb	381.31	K	Joback Method
tc	574.19	K	Joback Method
tf	201.80	K	Joback Method
vc	0.419	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	202.29	J/mol×K	381.31	Joback Method
cpg	217.80	J/mol×K	413.46	Joback Method
cpg	232.19	J/mol×K	445.60	Joback Method
cpg	245.53	J/mol×K	477.75	Joback Method
cpg	257.93	J/mol×K	509.90	Joback Method
cpg	269.48	J/mol×K	542.04	Joback Method
cpg	280.27	J/mol×K	574.19	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C1121353&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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