

# Benzene, 1,2-dichloro-4-methyl-

|                             |   |
|-----------------------------|---|
| <b>Other names:</b>         | 3,4-Dichlorotoluene<br>Toluene, 3,4-dichloro-   |
| <b>Inchi:</b>               | InChI=1S/C7H6Cl2/c1-5-2-3-6(8)7(9)4-5/h2-4H,1H3 |
| <b>InchiKey:</b>            | WYUIWKFIFOJVKW-UHFFFAOYSA-N                     |
| <b>Formula:</b>             | C7H6Cl2   |
| <b>SMILES:</b>              | Cc1ccc(Cl)c(Cl)c1                               |
| <b>Mol. weight [g/mol]:</b> | 161.03  |
| <b>CAS:</b>                 | 95-75-0   |

## Physical Properties

| Property code | Value         | Unit                 | Source         |
|---------------|---------------|----------------------|----------------|
| gf            | 77.35         | kJ/mol               | Joback Method  |
| hf            | -5.70         | kJ/mol               | Joback Method  |
| hfus          | 15.54         | kJ/mol               | Joback Method  |
| hvap          | 43.55         | kJ/mol               | Joback Method  |
| ie            | 8.85          | eV                   | NIST Webbook   |
| log10ws       | -3.32         |                      | Crippen Method |
| logp          | 3.302         |                      | Crippen Method |
| mcpol         | 110.210       | ml/mol               | McGowan Method |
| pc            | 3615.89       | kPa                  | Joback Method  |
| rinpol        | 1125.00       |                      | NIST Webbook   |
| rinpol        | 1126.30       |                      | NIST Webbook   |
| rinpol        | 1139.00       |                      | NIST Webbook   |
| tb            | 482.10        | K                    | NIST Webbook   |
| tc            | 700.48        | K                    | Joback Method  |
| tf            | 257.90 ± 0.02 | K                    | NIST Webbook   |
| vc            | 0.417         | m <sup>3</sup> /kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 186.67 | J/mol×K | 471.06          | Joback Method |
| cpg           | 195.96 | J/mol×K | 509.30          | Joback Method |
| cpg           | 204.68 | J/mol×K | 547.53          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 212.85    | J/molxK | 585.77 | Joback Method |
| cpg   | 220.49    | J/molxK | 624.01 | Joback Method |
| cpg   | 227.64    | J/molxK | 662.24 | Joback Method |
| cpg   | 234.30    | J/molxK | 700.48 | Joback Method |
| dvisc | 0.0016666 | Paxs    | 279.95 | Joback Method |
| dvisc | 0.0010588 | Paxs    | 311.80 | Joback Method |
| dvisc | 0.0007317 | Paxs    | 343.65 | Joback Method |
| dvisc | 0.0005384 | Paxs    | 375.50 | Joback Method |
| dvisc | 0.0004156 | Paxs    | 407.36 | Joback Method |
| dvisc | 0.0003331 | Paxs    | 439.21 | Joback Method |
| dvisc | 0.0002751 | Paxs    | 471.06 | Joback Method |
| hvapt | 49.40     | kJ/mol  | 460.50 | NIST Webbook  |

## Pressure Dependent Properties

| Property code | Value  | Unit | Pressure [kPa] | Source       |
|---------------|--------|------|----------------|--------------|
| tbrp          | 473.70 | K    | 98.80          | NIST Webbook |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.39535e+01                   |
| Coeff. B                    | -3.79884e+03                  |
| Coeff. C                    | -7.51600e+01                  |
| Temperature range (K), min. | 353.14                        |
| Temperature range (K), max. | 514.74                        |

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C95750&Units=SI>

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>hvapt:</b>   | Enthalpy of vaporization at a given temperature |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>pvap:</b>    | Vapor pressure                                  |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tbrp:</b>    | Boiling point at reduced pressure               |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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