

# Naphthalene, 1,2-dimethyl-

<b>Other names:</b>	1,2-Dimethylnaphthalene
<b>Inchi:</b>	InChI=1S/C12H12/c1-9-7-8-11-5-3-4-6-12(11)10(9)2/h3-8H,1-2H3
<b>InchiKey:</b>	QNLZIZAQLLYXTC-UHFFFAOYSA-N
<b>Formula:</b>	C12H12
<b>SMILES:</b>	Cc1ccc2ccccc2c1C
<b>Mol. weight [g/mol]:</b>	156.22
<b>CAS:</b>	573-98-8

## Physical Properties

Property code	Value	Unit	Source
af	0.4430		KDB
gf	249.96	kJ/mol	Joback Method
hf	113.65	kJ/mol	Joback Method
hfus	17.12	kJ/mol	Joback Method
hvap	47.55	kJ/mol	Joback Method
log10ws	-4.23		Crippen Method
logp	3.457		Crippen Method
mcvol	136.720	ml/mol	McGowan Method
pc	3010.00	kPa	KDB
rinpol	1440.00		NIST Webbook
rinpol	1448.00		NIST Webbook
rinpol	1447.00		NIST Webbook
rinpol	1456.00		NIST Webbook
rinpol	1430.00		NIST Webbook
rinpol	1447.00		NIST Webbook
rinpol	1439.70		NIST Webbook
rinpol	1455.10		NIST Webbook
rinpol	1432.00		NIST Webbook
rinpol	1452.20		NIST Webbook
rinpol	1451.10		NIST Webbook
rinpol	1439.60		NIST Webbook
rinpol	1452.20		NIST Webbook
rinpol	1459.10		NIST Webbook
rinpol	1439.60		NIST Webbook
rinpol	1452.20		NIST Webbook
rinpol	1459.10		NIST Webbook
rinpol	1452.20		NIST Webbook

rinpol	1451.10		NIST Webbook
rinpol	1462.00		NIST Webbook
rinpol	1450.60		NIST Webbook
rinpol	1439.00		NIST Webbook
rinpol	1440.00		NIST Webbook
rinpol	1444.00		NIST Webbook
rinpol	1433.00		NIST Webbook
rinpol	1432.00		NIST Webbook
rinpol	1432.00		NIST Webbook
rinpol	248.92		NIST Webbook
rinpol	236.49		NIST Webbook
rinpol	249.53		NIST Webbook
rinpol	1420.20		NIST Webbook
rinpol	247.14		NIST Webbook
rinpol	249.02		NIST Webbook
rinpol	248.47		NIST Webbook
rinpol	248.24		NIST Webbook
rinpol	249.04		NIST Webbook
rinpol	248.70		NIST Webbook
rinpol	249.67		NIST Webbook
rinpol	248.50		NIST Webbook
rinpol	246.49		NIST Webbook
rinpol	247.30		NIST Webbook
rinpol	246.50		NIST Webbook
rinpol	249.70		NIST Webbook
rinpol	246.49		NIST Webbook
rinpol	248.60		NIST Webbook
rinpol	248.80		NIST Webbook
rinpol	248.80		NIST Webbook
rinpol	1432.00		NIST Webbook
rinpol	1433.00		NIST Webbook
rinpol	247.30		NIST Webbook
rinpol	248.80		NIST Webbook
rinpol	1420.20		NIST Webbook
rinpol	1430.00		NIST Webbook
rinpol	1447.00		NIST Webbook
rinpol	246.56		NIST Webbook
ripol	2073.00		NIST Webbook
ripol	2072.00		NIST Webbook
ripol	2072.00		NIST Webbook
ripol	2072.00		NIST Webbook
tb	539.70	K	NIST Webbook
tb	498.00 ± 30.00	K	NIST Webbook
tb	541.70 ± 2.00	K	NIST Webbook

tb	539.00 ± 6.00	K	NIST Webbook
tb	539.50	K	KDB
tc	775.30	K	KDB
tf	272.00	K	KDB
tf	270.15 ± 0.70	K	NIST Webbook
vc	0.521	m <sup>3</sup> /kmol	KDB
zc	0.2435090		KDB

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	371.48	J/mol×K	761.14	Joback Method
cpg	296.31	J/mol×K	529.58	Joback Method
cpg	311.16	J/mol×K	568.17	Joback Method
cpg	325.00	J/mol×K	606.77	Joback Method
cpg	337.89	J/mol×K	645.36	Joback Method
cpg	349.89	J/mol×K	683.95	Joback Method
cpg	361.06	J/mol×K	722.55	Joback Method
dvisc	0.0002960	Paxs	529.58	Joback Method
dvisc	0.0012561	Paxs	309.16	Joback Method
dvisc	0.0008686	Paxs	345.90	Joback Method
dvisc	0.0006448	Paxs	382.63	Joback Method
dvisc	0.0005043	Paxs	419.37	Joback Method
dvisc	0.0004103	Paxs	456.11	Joback Method
dvisc	0.0003443	Paxs	492.84	Joback Method
hvapt	50.21	kJ/mol	539.50	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.43726e+01
Coeff. B	-4.33355e+03
Coeff. C	-9.54250e+01
Temperature range (K), min.	403.10
Temperature range (K), max.	573.69

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	1.39506e+02
Coeff. B	-1.36972e+04
Coeff. C	-1.77204e+01
Coeff. D	6.76958e-06
Temperature range (K), min.	402.15
Temperature range (K), max.	775.34

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=770">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=770</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C573988&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C573988&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=770">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=770</a>

## Legend

<b>af:</b>	Acentric Factor
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume
<b>zc:</b>	Critical Compressibility

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