

1,8-Dichlorooctane

Other names:	Octane, 1,8-dichloro-
Inchi:	InChI=1S/C8H16Cl2/c9-7-5-3-1-2-4-6-8-10/h1-8H2
InchiKey:	WXYMNDFVLNUAIA-UHFFFAOYSA-N
Formula:	C8H16Cl2
SMILES:	C1CCCCCCCCI
Mol. weight [g/mol]:	183.12
CAS:	2162-99-4

Physical Properties

Property code	Value	Unit	Source
gf	-7.38	kJ/mol	Joback Method
hf	-239.93	kJ/mol	Joback Method
hfus	24.87	kJ/mol	Joback Method
hvap	65.60	kJ/mol	NIST Webbook
log10ws	-3.48		Crippen Method
logp	3.805		Crippen Method
mcvol	148.060	ml/mol	McGowan Method
pc	2356.49	kPa	Joback Method
rinpol	1332.00		NIST Webbook
rinpol	1332.00		NIST Webbook
tb	514.20	K	NIST Webbook
tc	634.68	K	Joback Method
tf	239.76	K	Joback Method
vc	0.582	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	348.84	J/molxK	605.12	Joback Method
cpg	338.48	J/molxK	575.55	Joback Method
cpg	327.65	J/molxK	545.99	Joback Method
cpg	316.32	J/molxK	516.43	Joback Method
cpg	304.47	J/molxK	486.86	Joback Method
cpg	292.09	J/molxK	457.30	Joback Method

cpg	358.72	J/molxK	634.68	Joback Method
dvisc	0.0047665	Paxs	239.76	Joback Method
dvisc	0.0002927	Paxs	457.30	Joback Method
dvisc	0.0003814	Paxs	421.04	Joback Method
dvisc	0.0005225	Paxs	384.79	Joback Method
dvisc	0.0007642	Paxs	348.53	Joback Method
dvisc	0.0012208	Paxs	312.27	Joback Method
dvisc	0.0022058	Paxs	276.02	Joback Method
hvapt	55.90	kJ/mol	460.00	NIST Webbook
hvapt	47.30	kJ/mol	415.50	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	388.70	K	1.50	NIST Webbook
tbrp	391.70	K	1.90	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.39451e+01
Coeff. B	-4.07901e+03
Coeff. C	-7.68560e+01
Temperature range (K), min.	375.52
Temperature range (K), max.	549.31

Sources

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2162994&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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