

Fluoren-9-one, 2-ethylamino-

Inchi:	InChI=1S/C15H13NO/c1-2-16-10-7-8-12-11-5-3-4-6-13(11)15(17)14(12)9-10/h3-9,16H,2
InchiKey:	BQSDZCAYCJKMMP-UHFFFAOYSA-N
Formula:	C15H13NO
SMILES:	CCNc1ccc2c(c1)C(=O)c1ccccc1-2
Mol. weight [g/mol]:	223.27
CAS:	207291-46-1

Physical Properties

Property code	Value	Unit	Source
gf	330.81	kJ/mol	Joback Method
hf	106.95	kJ/mol	Joback Method
hfus	27.39	kJ/mol	Joback Method
hvap	66.08	kJ/mol	Joback Method
log10ws	-4.71		Crippen Method
logp	3.330		Crippen Method
mcvol	175.380	ml/mol	McGowan Method
pc	2829.33	kPa	Joback Method
tb	731.76	K	Joback Method
tc	978.56	K	Joback Method
tf	499.31	K	Joback Method
vc	0.675	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	476.49	J/mol×K	731.76	Joback Method
cpg	490.53	J/mol×K	772.89	Joback Method
cpg	503.57	J/mol×K	814.03	Joback Method
cpg	515.70	J/mol×K	855.16	Joback Method
cpg	527.01	J/mol×K	896.29	Joback Method
cpg	537.61	J/mol×K	937.42	Joback Method
cpg	547.58	J/mol×K	978.56	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C207291461&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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