

# Ethyl 3,4-dimethoxyphenyl acetate

<b>Other names:</b>	Benzeneacetic acid, 3,4-dimethoxy-, ethyl ester Acetic acid, (3,4-dimethoxyphenyl)-, ethyl ester Ethyl dimethoxyphenylacetate Homoveratric acid, ethyl ester
<b>Inchi:</b>	InChI=1S/C12H16O4/c1-4-16-12(13)8-9-5-6-10(14-2)11(7-9)15-3/h5-7H,4,8H2,1-3H3
<b>InchiKey:</b>	WZKCZNJTDZCNMH-UHFFFAOYSA-N
<b>Formula:</b>	C12H16O4
<b>SMILES:</b>	CCOC(=O)Cc1ccc(OC)c(OC)c1
<b>Mol. weight [g/mol]:</b>	224.25
<b>CAS:</b>	18066-68-7

## Physical Properties

Property code	Value	Unit	Source
gf	-300.61	kJ/mol	Joback Method
hf	-586.66	kJ/mol	Joback Method
hfus	25.26	kJ/mol	Joback Method
hvap	59.88	kJ/mol	Joback Method
log10ws	-2.21		Crippen Method
logp	1.809		Crippen Method
mcvol	175.360	ml/mol	McGowan Method
pc	2363.37	kPa	Joback Method
tb	631.73	K	Joback Method
tc	835.57	K	Joback Method
tf	393.08	K	Joback Method
vc	0.659	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	444.38	J/molxK	631.73	Joback Method
cpg	458.60	J/molxK	665.70	Joback Method
cpg	472.11	J/molxK	699.68	Joback Method
cpg	484.90	J/molxK	733.65	Joback Method
cpg	496.95	J/molxK	767.63	Joback Method

cpg	508.24	J/mol×K	801.60	Joback Method
cpg	518.77	J/mol×K	835.57	Joback Method
dvisc	0.0007650	Paxs	393.08	Joback Method
dvisc	0.0004811	Paxs	432.86	Joback Method
dvisc	0.0003272	Paxs	472.63	Joback Method
dvisc	0.0002362	Paxs	512.40	Joback Method
dvisc	0.0001787	Paxs	552.18	Joback Method
dvisc	0.0001404	Paxs	591.96	Joback Method
dvisc	0.0001137	Paxs	631.73	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18066687&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18066687&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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