

# 2-Amino-5-methylbenzoic acid

<b>Other names:</b>	5-Methylantranilic acid Benzoic acid, 2-amino-5-methyl- m-Toluic acid, 6-amino- 6-amino-m-toluic acid
<b>Inchi:</b>	InChI=1S/C8H9NO2/c1-5-2-3-7(9)6(4-5)8(10)11/h2-4H,9H2,1H3,(H,10,11)
<b>InchiKey:</b>	NBUUUJWWOARGNW-UHFFFAOYSA-N
<b>Formula:</b>	C8H9NO2
<b>SMILES:</b>	<chem>Cc1ccc(N)c(C(=O)O)c1</chem>
<b>Mol. weight [g/mol]:</b>	151.16
<b>CAS:</b>	2941-78-8

## Physical Properties

Property code	Value	Unit	Source
gf	-89.66	kJ/mol	Joback Method
hf	-225.88	kJ/mol	Joback Method
h <sub>fus</sub>	20.62	kJ/mol	Joback Method
h <sub>sub</sub>	110.60 ± 1.90	kJ/mol	NIST Webbook
h <sub>vap</sub>	71.07	kJ/mol	Joback Method
log <sub>10</sub> w <sub>s</sub>	-1.64		Crippen Method
log <sub>p</sub>	1.275		Crippen Method
m <sub>cvol</sub>	117.240	ml/mol	McGowan Method
pc	4608.87	kPa	Joback Method
tb	637.66	K	Joback Method
tc	853.80	K	Joback Method
tf	425.39	K	Joback Method
vc	0.429	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	284.29	J/mol×K	637.66	Joback Method
c <sub>pg</sub>	293.26	J/mol×K	673.68	Joback Method
c <sub>pg</sub>	301.67	J/mol×K	709.71	Joback Method
c <sub>pg</sub>	309.53	J/mol×K	745.73	Joback Method

cpg	316.86	J/mol×K	781.76	Joback Method
cpg	323.68	J/mol×K	817.78	Joback Method
cpg	330.01	J/mol×K	853.80	Joback Method
hfust	27.63	kJ/mol	450.00	NIST Webbook
hsubt	108.90 ± 0.50	kJ/mol	353.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2941788&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2941788&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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