

# Butane, 1-chloro-2-methyl-, (S)-

<b>Inchi:</b>	InChI=1S/C5H11Cl/c1-3-5(2)4-6/h5H,3-4H2,1-2H3/t5-/m1/s1
<b>InchiKey:</b>	IWAKWOFEHSYKSI-RXMQYKEDSA-N
<b>Formula:</b>	C5H11Cl
<b>SMILES:</b>	CCC(C)CCl
<b>Mol. weight [g/mol]:</b>	106.59
<b>CAS:</b>	40560-29-0

## Physical Properties

Property code	Value	Unit	Source
gf	-23.15	kJ/mol	Joback Method
hf	-167.55	kJ/mol	Joback Method
hfus	9.38	kJ/mol	Joback Method
hvap	30.72	kJ/mol	Joback Method
log10ws	-1.83		Crippen Method
logp	2.271		Crippen Method
mvol	93.550	ml/mol	McGowan Method
pc	3352.86	kPa	Joback Method
tb	372.70 ± 3.00	K	NIST Webbook
tb	372.65 ± 1.00	K	NIST Webbook
tc	527.49	K	Joback Method
tf	161.03	K	Joback Method
vc	0.358	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	150.99	J/mol×K	350.79	Joback Method
cpg	193.96	J/mol×K	498.04	Joback Method
cpg	186.05	J/mol×K	468.59	Joback Method
cpg	177.80	J/mol×K	439.14	Joback Method
cpg	169.22	J/mol×K	409.69	Joback Method
cpg	160.28	J/mol×K	380.24	Joback Method
cpg	201.54	J/mol×K	527.49	Joback Method
dvisc	0.0002823	Paxs	350.79	Joback Method

dvisc	0.0003734	Paxs	319.16	Joback Method
dvisc	0.0005253	Paxs	287.54	Joback Method
dvisc	0.0008040	Paxs	255.91	Joback Method
dvisc	0.0013876	Paxs	224.28	Joback Method
dvisc	0.0028647	Paxs	192.66	Joback Method
dvisc	0.0078623	Paxs	161.03	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C40560290&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C40560290&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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