

Sulfuric Acid

Other names:	Acide sulfurique
	Acido solforico
	BOV
	Battery acid
	Dihydrogen sulfate
	Dipping acid
	Electrolyte acid
	H2SO4
	HYDROGEN SULFATE
	Matting acid
	Mattling acid
	Nordhausen acid
	O2S(OH)2
	Oil of vitriol
	SULPHURIC ACID
	Schwefelsaeureloesungen
	Spirit of alum
	Spirit of vitriol
	Vitriol brown oil
	Vitriol, oil of
	Zwavelzuuroplossingen
Inchi:	InChI=1S/H2O4S/c1-5(2,3)4/h(H2,1,2,3,4)
InchiKey:	QAOWNCQODCNURD-UHFFFAOYSA-N
Formula:	H2O4S
SMILES:	O=S(=O)(O)O
Mol. weight [g/mol]:	98.08
CAS:	7664-93-9

Physical Properties

Property code	Value	Unit	Source
affp	699.40	kJ/mol	NIST Webbook
basg	666.90	kJ/mol	NIST Webbook
basg	683.00 ± 5.00	kJ/mol	NIST Webbook
basg	680.00	kJ/mol	NIST Webbook
ep	4.90	J/mol×K	NIST Webbook
gf	-793.06	kJ/mol	Joback Method

hf	-801.14	kJ/mol	Joback Method
hfpi	500.00 ± 20.00	kJ/mol	NIST Webbook
hfus	15.31	kJ/mol	Joback Method
hvap	67.59	kJ/mol	Joback Method
ie	12.40 ± 0.05	eV	NIST Webbook
log10ws	0.72		Crippen Method
logp	-0.653		Crippen Method
mcvol	50.690	ml/mol	McGowan Method
pc	12942.62	kPa	Joback Method
tb	431.54	K	Joback Method
tc	590.76	K	Joback Method
tf	249.96	K	Joback Method
vc	0.200	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	93.11	J/mol×K	431.54	Joback Method
cpg	95.72	J/mol×K	458.08	Joback Method
cpg	98.26	J/mol×K	484.61	Joback Method
cpg	100.73	J/mol×K	511.15	Joback Method
cpg	103.13	J/mol×K	537.68	Joback Method
cpg	105.43	J/mol×K	564.22	Joback Method
cpg	107.65	J/mol×K	590.76	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.70628e+01
Coeff. B	-6.85442e+03
Coeff. C	-5.92000e+01
Temperature range (K), min.	298.15
Temperature range (K), max.	610.00

Sources

- Revisiting dibenzothioephene**: thermochemical data: Experimental and computational study: Modeling of MgSO₄*7H₂O in the TiO₂-ZrO₂-H₂SIO₃-H₂O-Cu(II) complexes with N,N-diethylurea and correlation for the solubility of various hydrophilic ureas: A thermodynamic analysis of H₂SO₄+H₂O and MgSO₄.H₂SO₄.H₂O Systems Methodology: Paracetic acid in aqueous sulfuric acid solutions: Energetic study of benzothiazole and two methylbenzothiazole derivatives: Determination and optimization modeling of the solubility of FeSO₄*7H₂O in the First Law composition of two bis(N,N-diethylothioureas): Experimental and computational thermochemical study of benzofuran, benzothiophene and indole derivatives:
- Thermochemical and thermophysical study of 2-thiophenecarboxylic acid** Perfluorinated aliphatic carboxylic acids Thermophysical Properties of a Sulfuric Acid-Water Mixture: Experimental and Theoretical Study of Liquid-Liquid Equilibria of Nitrobenzene-Inorganic Acids Systems Relative Densities and Viscosities of: Sulfuric Acid + Dimethylamine + Water Thermodynamic properties of epsilon-caprolactam and epsilon-caprolactam-H group in gaseous phase: Experimental and theoretical energies and formation enthalpies of 2-SH-benzazoles: Thermodynamic study of 2-aminothiazole and 2-mercaptobenzothiazole: Experimental results at medium pressures and determination of related systems' solubility data (I) in the preparation process of K₂SO₄/SiO₂/P₂O₅/Copolymer/Water Binary Mixtures from T) (298.15 to 330.15K) triadiazole: study of its structure, energetics and aromaticity: Enthalpies of combustion and formation of 2-acetylpyrrole, Redesigning the rotating bomb combustion calorimeter: Thermochemical properties of three 2-thiophenecarboxylic acid derivatives: Sulfur-Containing Ionic Liquids. Rotating-Bomb Combustion Calorimetry Quantifies the Formation and Sublimation of Methoxyphenylamines and 2-Thienepyrroles Comprehensive Thermochemical Study of Cyclic Five- and Six-Membered Nitriles and/or Viscosities of Nicotin + 3-Picoline + Sulfuric Acid + Water from Calculated Data: Study of methyl and ethyl 2-thiophenecarboxylates and ethyl 2-methyl-2-thiophenacetates:
- Investigation of electrolytes of the vanadium redox flow battery (VII)**: Experimental study on the mixed thermoelectric system of ZnS/CdTe/H₂SO₄ + SnCl₄ as the electrolyte: Synthesis and characterization of 6-Amino-2-thiopyranose in the crystallization and conformational studies of optical active phenylbenzazole derivatives barbituric and 2-thiobarbituric acids: An experimental and computational study:
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Experimental and computational thermochemical studies of 6-azauracil vapor-liquid Equilibria of the FeSO_4 $\text{Fe}_2(\text{SO}_4)_3$ H_2SO_4 H_2O System at (30, 60, 90, and 101.3) kPa:

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Legend

affp:	Proton affinity
basg:	Gas basicity
cpg:	Ideal gas heat capacity
ep:	Protonation entropy at 298K
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfpi:	Enthalpy of formation of positive ion at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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