## **D-Alanine**

Other names: (S)-(+)-alanine

(S)-2-aminopropanoic acid

.alpha.-alanine Alanine, D-Ba 2776

D(-)-«alpha»-Alanine

D-(-)-Alanine

D-«alpha»-Alanine

L-.alpha.-aminopropionic acid L-2-aminopropanoic acid

L-alanine

InChl=1S/C3H7NO2/c1-2(4)3(5)6/h2H,4H2,1H3,(H,5,6)/t2-/m0/s1

InchiKey: QNAYBMKLOCPYGJ-REOHCLBHSA-N

Formula: C3H7NO2 SMILES: CC(N)C(=O)O

**Mol. weight [g/mol]:** 89.09 **CAS:** 338-69-2

# **Physical Properties**

| Property code | Value           | Unit    | Source         |  |
|---------------|-----------------|---------|----------------|--|
| chs           | -1623.00 ± 0.20 | kJ/mol  | NIST Webbook   |  |
| chs           | -1576.00 ± 3.50 | kJ/mol  | NIST Webbook   |  |
| chs           | -1619.60 ± 0.54 | kJ/mol  | NIST Webbook   |  |
| chs           | -1639.90        | kJ/mol  | NIST Webbook   |  |
| gf            | -227.35         | kJ/mol  | Joback Method  |  |
| hf            | -341.55         | kJ/mol  | Joback Method  |  |
| hfs           | -561.24 ± 0.59  | kJ/mol  | NIST Webbook   |  |
| hfs           | -605.00 ± 3.50  | kJ/mol  | NIST Webbook   |  |
| hfus          | 10.89           | kJ/mol  | Joback Method  |  |
| hvap          | 55.95           | kJ/mol  | Joback Method  |  |
| log10ws       | 0.28            |         | Crippen Method |  |
| logp          | -0.582          |         | Crippen Method |  |
| mcvol         | 70.550          | ml/mol  | McGowan Method |  |
| рс            | 6046.69         | kPa     | Joback Method  |  |
| SS            | 132.20          | J/mol×K | NIST Webbook   |  |
| tb            | 486.18          | K       | Joback Method  |  |
| tc            | 677.88          | K       | Joback Method  |  |

| tf | 302.58 | K       | Joback Method |
|----|--------|---------|---------------|
| VC | 0.252  | m3/kmol | Joback Method |

## **Temperature Dependent Properties**

| Property code | Value         | Unit    | Temperature [K] | Source        |
|---------------|---------------|---------|-----------------|---------------|
| cpg           | 151.30        | J/mol×K | 486.18          | Joback Method |
| cpg           | 157.40        | J/mol×K | 518.13          | Joback Method |
| cpg           | 163.22        | J/mol×K | 550.08          | Joback Method |
| cpg           | 168.75        | J/mol×K | 582.03          | Joback Method |
| cpg           | 174.01        | J/mol×K | 613.98          | Joback Method |
| cpg           | 179.00        | J/mol×K | 645.93          | Joback Method |
| cpg           | 183.73        | J/mol×K | 677.88          | Joback Method |
| cps           | 120.80        | J/mol×K | 296.80          | NIST Webbook  |
| hsubt         | 132.80        | kJ/mol  | 416.50          | NIST Webbook  |
| hsubt         | 138.00 ± 8.00 | kJ/mol  | 461.00          | NIST Webbook  |

### Sources

### McGowan Method:

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Conductometric analysis of 1-butyl-3-methylimidazolium ibuprofenate as an active pharmaceutical ingredient ionic liquid (API-IL) in the aqueous amino acids

**Legend** 

**chs:** Standard solid enthalpy of combustion

cpg: Ideal gas heat capacitycps: Solid phase heat capacity

gf: Standard Gibbs free energy of formationhf: Enthalpy of formation at standard conditions

**hfs:** Solid phase enthalpy of formation at standard conditions

**hfus:** Enthalpy of fusion at standard conditions

**hsubt:** Enthalpy of sublimation at a given temperature **hvap:** Enthalpy of vaporization at standard conditions

log10ws: Log10 of Water solubility in mol/llogp: Octanol/Water partition coefficientmcvol: McGowan's characteristic volume

pc: Critical Pressure

ss: Solid phase molar entropy at standard conditions

**tb:** Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) point

vc: Critical Volume

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