

# Baclofen

## Other names:

(. +/-)-Baclofen  
(. +/-)-Baklofen  
4-Amino-3-(4-chlorophenyl)butyric acid  
4-Amino-3-(p-chlorophenyl)butyric acid  
Ba 34647  
Baclon  
Benzenepropanoic acid, «beta»-(aminomethyl)-4-chloro-  
Benzenepropanoic acid, Â«betaÂ»-(aminomethyl)-4-chloro-  
Butanoic acid, 4-amino-3-(4-chlorophenyl)-  
C 34647Ba  
CIBA 34,647-Ba  
DL-4-Amino-3-p-chlorophenylbutanoic acid  
DL-Baclofen  
Hydrocinnamic acid, «beta»-(aminomethyl)-p-chloro-  
Hydrocinnamic acid, Â«betaÂ»-(aminomethyl)-p-chloro-  
Lioresal  
«beta»-(4-Chlorophenyl)gaba  
«beta»-(Aminomethyl)-4-chlorobenzenepropanoic acid  
«beta»-(Aminomethyl)-p-chlorohydrocinnamic acid  
«beta»-(p-Chlorophenyl)-«gamma»-aminobutyric acid  
«gamma»-Amino-«beta»-(p-chlorophenyl)butyric acid  
Â«betaÂ»-(4-Chlorophenyl)gaba  
Â«betaÂ»-(Aminomethyl)-4-chlorobenzenepropanoic acid  
Â«betaÂ»-(Aminomethyl)-p-chlorohydrocinnamic acid  
Â«betaÂ»-(p-Chlorophenyl)-Â«gammaÂ»-aminobutyric acid  
Â«gammaÂ»-Amino-Â«betaÂ»-(p-chlorophenyl)butyric acid

**Inchi:** InChI=1S/C10H12ClNO2/c11-9-3-1-7(2-4-9)8(6-12)5-10(13)14/h1-4,8H,5-6,12H2,(H,13,14)H3

**InchiKey:** KPYSYYIEGFHWSV-UHFFFAOYSA-N

**Formula:** C10H12ClNO2

**SMILES:** NCC(CC(=O)O)c1ccc(Cl)cc1

**Mol. weight [g/mol]:** 213.66

**CAS:** 1134-47-0

## Physical Properties

Property code	Value	Unit	Source
gf	-77.56	kJ/mol	Joback Method

hf	-276.71		kJ/mol	Joback Method
hfus	26.87		kJ/mol	Joback Method
hvap	78.86		kJ/mol	Joback Method
log10ws	-1.70			Aqueous and cosolvent solubility data for drug-like organic compounds
log10ws	-1.78			Aqueous Solubility Prediction Method
logp	1.857			Crippen Method
mcvol	157.660		ml/mol	McGowan Method
pc	3547.31		kPa	Joback Method
rinpola	2018.10			NIST Webbook
rinpola	2018.10			NIST Webbook
tb	715.43		K	Joback Method
tc	931.06		K	Joback Method
tf	480.48		K	Aqueous Solubility Prediction Method
vc	0.585		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	404.48	J/mol×K	715.43	Joback Method
cpg	414.47	J/mol×K	751.37	Joback Method
cpg	423.74	J/mol×K	787.31	Joback Method
cpg	432.35	J/mol×K	823.24	Joback Method
cpg	440.32	J/mol×K	859.18	Joback Method
cpg	447.70	J/mol×K	895.12	Joback Method
cpg	454.51	J/mol×K	931.06	Joback Method

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousD>

**Aqueous and cosolvent solubility data for drug-like organic compounds:** <https://www.ncbi.nlm.nih.gov/pmc/articles/PMC2751500/>

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C1134470&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307i>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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