

# 3,5-Dimethyl-4-heptanone

<b>Other names:</b>	3,5-dimethylheptan-4-one 4-Heptanone, 3,5-dimethyl-
<b>Inchi:</b>	InChI=1S/C9H18O/c1-5-7(3)9(10)8(4)6-2/h7-8H,5-6H2,1-4H3
<b>InchiKey:</b>	VZXXYILNWWRSGE-UHFFFAOYSA-N
<b>Formula:</b>	C9H18O
<b>SMILES:</b>	CCC(C)C(=O)C(C)CC
<b>Mol. weight [g/mol]:</b>	142.24
<b>CAS:</b>	19549-84-9

## Physical Properties

Property code	Value	Unit	Source
gf	-108.90	kJ/mol	Joback Method
hf	-352.23	kJ/mol	Joback Method
hfus	13.62	kJ/mol	Joback Method
hvap	41.60	kJ/mol	Joback Method
log10ws	-2.39		Crippen Method
logp	2.648		Crippen Method
mcvol	139.240	ml/mol	McGowan Method
pc	2492.52	kPa	Joback Method
tb	435.00 ± 4.00	K	NIST Webbook
tb	435.20	K	NIST Webbook
tc	640.10	K	Joback Method
tf	211.12	K	Joback Method
vc	0.533	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	297.92	J/mol×K	458.31	Joback Method
cpg	312.19	J/mol×K	488.61	Joback Method
cpg	325.86	J/mol×K	518.91	Joback Method
cpg	338.94	J/mol×K	549.20	Joback Method
cpg	351.45	J/mol×K	579.50	Joback Method
cpg	363.40	J/mol×K	609.80	Joback Method

cpg	374.81	J/mol×K	640.10	Joback Method
dvisc	0.0112700	Paxs	211.12	Joback Method
dvisc	0.0035885	Paxs	252.32	Joback Method
dvisc	0.0015755	Paxs	293.52	Joback Method
dvisc	0.0008471	Paxs	334.72	Joback Method
dvisc	0.0005218	Paxs	375.91	Joback Method
dvisc	0.0003537	Paxs	417.11	Joback Method
dvisc	0.0002572	Paxs	458.31	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.50321e+01
Coeff. B	-3.88515e+03
Coeff. C	-6.21220e+01
Temperature range (K), min.	325.62
Temperature range (K), max.	461.80

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19549849&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19549849&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/24-917-7/3-5-Dimethyl-4-heptanone.pdf>

Generated by Cheméo on 2025-01-15 19:31:43.611275326 +0000 UTC m=+177719.458200957.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.