

# 2-Propenoic acid, 2-methyl-, 1,2-ethanediylbis(oxy-2,1-ethanediyl) ester

<b>Other names:</b>	(ethane-1,2-diylbis(oxy))bis(ethane-2,1-diyl) bis(2-methylacrylate) 1,2-Bis[2-(Methacryloyloxy)ethoxy]ethane 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate) 2,2'-ethylenedioxydiethyl dimethacrylate 2-Propenoic acid, 2-methyl-, 1,1'-[1,2-ethanediylbis(oxy-2,1-ethanediyl)] ester Bisomer TEGDMA Ethylenebis(oxyethylene) methacrylate Methacrylic acid, diester with triethylene glycol NK Ester 3G NSC 84260 Polyester TGM-3 TEDMA TGM 3 TGM 35 TGM 3PC TGM 3S Triethylene dimethacrylate ethane-1,2-diylbis(oxyethane-2,1-diyl) bis(2-methylacrylate) ethane-1,2-diylbis(oxyethane-2,1-diyl) bis(2-methylprop-2-enoate) triethylene glycol dimethacrylate
<b>Inchi:</b>	InChI=1S/C14H22O6/c1-11(2)13(15)19-9-7-17-5-6-18-8-10-20-14(16)12(3)4/h1,3,5-10H2
<b>InchiKey:</b>	HWSSEYVMGDIFMH-UHFFFAOYSA-N
<b>Formula:</b>	C14H22O6
<b>SMILES:</b>	C=C(C)C(=O)OCCOCCOCCOC(=O)C(=C)C
<b>Mol. weight [g/mol]:</b>	286.32
<b>CAS:</b>	109-16-0

## Physical Properties

Property code	Value	Unit	Source
gf	-452.26	kJ/mol	Joback Method
hf	-855.05	kJ/mol	Joback Method
hfus	34.79	kJ/mol	Joback Method
hvap	68.71	kJ/mol	Joback Method
log10ws	-1.29		Crippen Method
logp	1.258		Crippen Method
mcvol	226.140	ml/mol	McGowan Method
pc	1721.73	kPa	Joback Method

tb	710.26	K	Joback Method
tc	894.77	K	Joback Method
tf	404.88	K	Joback Method
vc	0.868	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	630.66	J/mol×K	710.26	Joback Method
cpg	645.14	J/mol×K	741.01	Joback Method
cpg	658.86	J/mol×K	771.76	Joback Method
cpg	671.80	J/mol×K	802.51	Joback Method
cpg	683.97	J/mol×K	833.26	Joback Method
cpg	695.35	J/mol×K	864.02	Joback Method
cpg	705.94	J/mol×K	894.77	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	444.20	K	0.70	NIST Webbook

## Sources

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C109160&Units=SI>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method: [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

High-pressure phase behavior of tri-ethylene glycol dimethacrylate and <https://www.doi.org/10.1016/j.fluid.2011.12.011>

Joback Method: [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m cvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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