

# 3-Hexenoic acid

<b>Other names:</b>	Hex-3-enoic acid Hydrosorbic acid
<b>Inchi:</b>	InChI=1S/C6H10O2/c1-2-3-4-5-6(7)8/h3-4H,2,5H2,1H3,(H,7,8)
<b>InchiKey:</b>	XXHDAWYDNSXJQM-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O2
<b>SMILES:</b>	CCC=CCC(=O)O
<b>Mol. weight [g/mol]:</b>	114.14
<b>CAS:</b>	4219-24-3

## Physical Properties

Property code	Value	Unit	Source
chl	-3338.00	kJ/mol	NIST Webbook
chl	-3342.10	kJ/mol	NIST Webbook
gf	-185.88	kJ/mol	Joback Method
hf	-314.76	kJ/mol	Joback Method
hfus	17.19	kJ/mol	Joback Method
hvap	52.33	kJ/mol	Joback Method
log10ws	-1.29		Crippen Method
logp	1.427		Crippen Method
mcvol	98.540	ml/mol	McGowan Method
pc	3970.51	kPa	Joback Method
rinpol	983.00		NIST Webbook
rinpol	1023.00		NIST Webbook
rinpol	995.00		NIST Webbook
rinpol	983.00		NIST Webbook
ripol	1954.00		NIST Webbook
ripol	1954.00		NIST Webbook
ripol	1977.00		NIST Webbook
ripol	1977.00		NIST Webbook
tb	481.20	K	NIST Webbook
tc	665.94	K	Joback Method
tf	263.05	K	Joback Method
vc	0.377	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	206.08	J/molxK	486.89	Joback Method
cpg	244.26	J/molxK	636.10	Joback Method
cpg	237.37	J/molxK	606.26	Joback Method
cpg	230.13	J/molxK	576.41	Joback Method
cpg	222.51	J/molxK	546.57	Joback Method
cpg	214.50	J/molxK	516.73	Joback Method
cpg	250.80	J/molxK	665.94	Joback Method
dvisc	0.0001689	Paxs	486.89	Joback Method
dvisc	0.0002731	Paxs	449.58	Joback Method
dvisc	0.0004814	Paxs	412.28	Joback Method
dvisc	0.0009502	Paxs	374.97	Joback Method
dvisc	0.0021795	Paxs	337.66	Joback Method
dvisc	0.0061439	Paxs	300.36	Joback Method
dvisc	0.0232383	Paxs	263.05	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.56843e+01
Coeff. B	-4.49600e+03
Coeff. C	-7.49100e+01
Temperature range (K), min.	366.92
Temperature range (K), max.	508.35

## Sources

- Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>
- Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)
- Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)
- McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>
- NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C4219243&Units=SI>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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