

# 1,2,4,5-Benzene-tetracarboxylic acid

<b>Other names:</b>	1,2,4,5-benzenetetracarboxylic acid 1,2,4,5-tetracarboxybenzene Benzene-1,2,4,5-tetracarboxylic acid USAF xr-20 pyromellitic acid
<b>Inchi:</b>	InChI=1S/C10H6O8/c11-7(12)3-1-4(8(13)14)6(10(17)18)2-5(3)9(15)16/h1-2H,(H,11,12)(H,13,14)
<b>InchiKey:</b>	CYIDZMCFTVVTJO-UHFFFAOYSA-N
<b>Formula:</b>	C10H6O8
<b>SMILES:</b>	O=C(O)c1cc(C(=O)O)c(C(=O)O)cc1C(=O)O
<b>Mol. weight [g/mol]:</b>	254.15
<b>CAS:</b>	89-05-4

## Physical Properties

Property code	Value	Unit	Source
chs	-3221.80 ± 0.88	kJ/mol	NIST Webbook
gf	-946.12	kJ/mol	Joback Method
hf	-1106.85	kJ/mol	Joback Method
hfs	-1570.80 ± 0.88	kJ/mol	NIST Webbook
hfus	37.28	kJ/mol	Joback Method
hvap	135.82	kJ/mol	Joback Method
log10ws	-1.71		Crippen Method
logp	0.479		Crippen Method
mvol	157.760	ml/mol	McGowan Method
pc	6180.53	kPa	Joback Method
tb	1054.02	K	Joback Method
tc	1292.89	K	Joback Method
tf	544.15 ± 2.00	K	NIST Webbook
vc	0.588	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	443.75	J/mol×K	1054.02	Joback Method
cpg	446.96	J/mol×K	1093.83	Joback Method

cpg	449.56	J/molxK	1133.64	Joback Method
cpg	451.57	J/molxK	1173.45	Joback Method
cpg	453.00	J/molxK	1213.26	Joback Method
cpg	453.89	J/molxK	1253.07	Joback Method
cpg	454.26	J/molxK	1292.89	Joback Method
dvisc	0.0000042	Paxs	709.44	Joback Method
dvisc	0.0000016	Paxs	766.87	Joback Method
dvisc	0.0000007	Paxs	824.30	Joback Method
dvisc	0.0000004	Paxs	881.73	Joback Method
dvisc	0.0000002	Paxs	939.16	Joback Method
dvisc	0.0000001	Paxs	996.59	Joback Method
dvisc	6.7288840e-08	Paxs	1054.02	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>The solubilities of benzene polycarboxylic acids in water:</b>	<a href="https://www.doi.org/10.1016/j.jct.2005.07.007">https://www.doi.org/10.1016/j.jct.2005.07.007</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C89054&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C89054&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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