

# 2-Butenal, 3-methyl-

<b>Other names:</b>	3,3-Dimethylacrolein 3-Methyl-2-butenal 3-Methyl-2-butenaldehyde 3-Methylbut-2-enal 3-Methylcrotonaldehyde 3-methyl-2-butenal (prenal) Crotonaldehyde, 3-methyl- NSC 149164 Prenal Senecialdehyde Senecioaldehyde «beta», «beta»-Dimethylacrolein «beta», «beta»-Dimethylacrylic aldehyde «beta»-Methylcrotonaldehyde Â«betaÂ», Â«betaÂ»-Dimethylacrolein Â«betaÂ», Â«betaÂ»-Dimethylacrylic aldehyde Â«betaÂ»-Methylcrotonaldehyde
<b>Inchi:</b>	InChI=1S/C5H8O/c1-5(2)3-4-6/h3-4H,1-2H3
<b>InchiKey:</b>	SEPQTYODOKLVSU-UHFFFAOYSA-N
<b>Formula:</b>	C5H8O
<b>SMILES:</b>	CC(C)=CC=O
<b>Mol. weight [g/mol]:</b>	84.12
<b>CAS:</b>	107-86-8

## Physical Properties

Property code	Value	Unit	Source
affp	856.90	kJ/mol	NIST Webbook
basg	825.00	kJ/mol	NIST Webbook
gf	-36.63	kJ/mol	Joback Method
hf	-124.68	kJ/mol	Joback Method
hfus	9.89	kJ/mol	Joback Method
hvap	33.48	kJ/mol	Joback Method
log10ws	-1.05		Crippen Method
logp	1.151		Crippen Method
mcvol	78.580	ml/mol	McGowan Method
pc	4072.51	kPa	Joback Method
rinpol	790.00		NIST Webbook

rinpol	800.00	NIST Webbook
rinpol	796.00	NIST Webbook
rinpol	758.00	NIST Webbook
rinpol	791.00	NIST Webbook
rinpol	753.00	NIST Webbook
rinpol	744.00	NIST Webbook
rinpol	791.00	NIST Webbook
rinpol	781.00	NIST Webbook
rinpol	790.00	NIST Webbook
rinpol	782.00	NIST Webbook
rinpol	737.00	NIST Webbook
rinpol	781.00	NIST Webbook
rinpol	753.00	NIST Webbook
rinpol	780.00	NIST Webbook
rinpol	791.00	NIST Webbook
rinpol	781.00	NIST Webbook
rinpol	782.00	NIST Webbook
rinpol	781.00	NIST Webbook
rinpol	780.00	NIST Webbook
rinpol	778.00	NIST Webbook
rinpol	755.00	NIST Webbook
rinpol	788.00	NIST Webbook
rinpol	787.00	NIST Webbook
rinpol	784.00	NIST Webbook
rinpol	791.00	NIST Webbook
rinpol	781.00	NIST Webbook
rinpol	783.30	NIST Webbook
rinpol	780.00	NIST Webbook
rinpol	784.00	NIST Webbook
rinpol	796.00	NIST Webbook
rinpol	748.40	NIST Webbook
rinpol	748.40	NIST Webbook
rinpol	748.40	NIST Webbook
ripol	1215.00	NIST Webbook
ripol	1212.00	NIST Webbook
ripol	1215.00	NIST Webbook
ripol	1200.00	NIST Webbook
ripol	1230.00	NIST Webbook
ripol	1221.00	NIST Webbook
ripol	1222.00	NIST Webbook
ripol	1220.50	NIST Webbook
ripol	1199.00	NIST Webbook
ripol	1233.00	NIST Webbook
ripol	1206.00	NIST Webbook

ripol	1236.00		NIST Webbook
ripol	1202.00		NIST Webbook
ripol	1199.00		NIST Webbook
ripol	1189.00		NIST Webbook
ripol	1202.00		NIST Webbook
ripol	1220.50		NIST Webbook
ripol	1236.00		NIST Webbook
ripol	1189.00		NIST Webbook
ripol	1215.00		NIST Webbook
tb	407.20	K	NIST Webbook
tc	552.54	K	Joback Method
tf	169.07	K	Joback Method
vc	0.314	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	130.05	J/mol×K	366.50	Joback Method
cpg	138.26	J/mol×K	397.51	Joback Method
cpg	146.07	J/mol×K	428.51	Joback Method
cpg	153.48	J/mol×K	459.52	Joback Method
cpg	160.51	J/mol×K	490.53	Joback Method
cpg	167.19	J/mol×K	521.53	Joback Method
cpg	173.52	J/mol×K	552.54	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.49139e+01
Coeff. B	-3.63291e+03
Coeff. C	-5.43380e+01
Temperature range (K), min.	302.72
Temperature range (K), max.	432.67

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C107868&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C107868&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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