

# Diazene, (4-chlorophenyl)phenyl-, (E)-

<b>Other names:</b>	Azobenzene, 4-chloro-, (E)- p-Chloro-trans-azobenzene
<b>Inchi:</b>	InChI=1S/C12H9ClN2/c13-10-6-8-12(9-7-10)15-14-11-4-2-1-3-5-11/h1-9H
<b>InchiKey:</b>	NJFDMENHTAYHMA-UHFFFAOYSA-N
<b>Formula:</b>	C12H9ClN2
<b>SMILES:</b>	Clc1ccc(N=Nc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	216.67
<b>CAS:</b>	6141-95-3

## Physical Properties

Property code	Value	Unit	Source
hf	202.06	kJ/mol	Joback Method
hvap	58.58	kJ/mol	Joback Method
ie	8.55 ± 0.05	eV	NIST Webbook
log10ws	-4.20		Crippen Method
logp	4.755		Crippen Method
mvol	160.320	ml/mol	McGowan Method
pc	2445.89	kPa	Joback Method
tb	718.93	K	Joback Method
tc	992.71	K	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6141953&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6141953&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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