

# Sebacic acid, 2-chloropropyl isobutyl ester

**Inchi:** InChI=1S/C17H31ClO4/c1-14(2)12-21-16(19)10-8-6-4-5-7-9-11-17(20)22-13-15(3)18/h14  
**InchiKey:** XJHDWUFLKOUYST-UHFFFAOYSA-N  
**Formula:** C17H31ClO4  
**SMILES:** CC(C)COC(=O)CCCCCCCCC(=O)OCC(C)Cl  
**Mol. weight [g/mol]:** 334.88

## Physical Properties

Property code	Value	Unit	Source
gf	-392.39	kJ/mol	Joback Method
hf	-910.11	kJ/mol	Joback Method
hfus	42.51	kJ/mol	Joback Method
hvap	75.36	kJ/mol	Joback Method
log10ws	-4.69		Crippen Method
logp	4.477		Crippen Method
mvol	277.510	ml/mol	McGowan Method
pc	1294.86	kPa	Joback Method
rinpol	2236.00		NIST Webbook
rinpol	2236.00		NIST Webbook
tb	777.49	K	Joback Method
tc	963.26	K	Joback Method
tf	425.59	K	Joback Method
vc	1.073	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	823.50	J/molxK	777.49	Joback Method
cpg	839.90	J/molxK	808.45	Joback Method
cpg	855.35	J/molxK	839.41	Joback Method
cpg	869.87	J/molxK	870.38	Joback Method
cpg	883.46	J/molxK	901.34	Joback Method
cpg	896.14	J/molxK	932.30	Joback Method
cpg	907.93	J/molxK	963.26	Joback Method
dvisc	0.0012997	Paxs	425.59	Joback Method

dvisc	0.0005702	Paxs	484.24	Joback Method
dvisc	0.0002990	Paxs	542.89	Joback Method
dvisc	0.0001778	Paxs	601.54	Joback Method
dvisc	0.0001159	Paxs	660.19	Joback Method
dvisc	0.0000811	Paxs	718.84	Joback Method
dvisc	0.0000598	Paxs	777.49	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U355443&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U355443&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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