

# Carbamic acid, methyl-, 3-methylphenyl ester

<b>Other names:</b>	3-Methylphenyl Methylcarbamate 3-Methylphenyl N-methylcarbamate 3-Tolyl N-methylcarbamate 3-Tolyl methylcarbamate Carbamic acid, N-methyl-, 3-methylphenyl ester Carbamic acid, methyl-, 3-tolyl ester Carbamic acid, methyl-, m-tolyl ester DRC 3341 Kumiai MTMC Metacrate Metholcarb Methylcarbamic acid m-tolyl ester Metolcarb NSC 91193 S 1065 Tsumacide Tsumaunka m-Cresyl N-methylcarbamate m-Cresyl ester of N-methylcarbamic acid m-Cresyl methylcarbamate m-Methylphenyl methylcarbamate m-Tolyl N-methylcarbamate m-Tolyl methylcarbamate m-Tolyvester kyseliny methylkarbaminove
<b>Inchi:</b>	InChI=1S/C9H11NO2/c1-7-4-3-5-8(6-7)12-9(11)10-2/h3-6H,1-2H3,(H,10,11)
<b>InchiKey:</b>	VOEYXMAFNDNED-UHFFFAOYSA-N
<b>Formula:</b>	C9H11NO2
<b>SMILES:</b>	CNC(=O)Oc1cccc(C)c1
<b>Mol. weight [g/mol]:</b>	165.19
<b>CAS:</b>	1129-41-5

## Physical Properties

Property code	Value	Unit	Source
gf	-16.85	kJ/mol	Joback Method
hf	-195.36	kJ/mol	Joback Method

hfus	20.60		kJ/mol	Joback Method
hvap	54.16		kJ/mol	Joback Method
log10ws	-1.80			Aqueous Solubility Prediction Method
logp	1.713			Crippen Method
mcvol	131.330		ml/mol	McGowan Method
pc	3431.89		kPa	Joback Method
rinpol	1465.00			NIST Webbook
rinpol	1392.00			NIST Webbook
rinpol	1465.00			NIST Webbook
rinpol	1411.00			NIST Webbook
ripol	2048.00			NIST Webbook
tb	563.44		K	Joback Method
tc	780.85		K	Joback Method
tf	349.65		K	Aqueous Solubility Prediction Method
vc	0.490		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	301.19	J/mol×K	563.44	Joback Method
cpg	313.64	J/mol×K	599.68	Joback Method
cpg	325.37	J/mol×K	635.91	Joback Method
cpg	336.41	J/mol×K	672.15	Joback Method
cpg	346.75	J/mol×K	708.38	Joback Method
cpg	356.42	J/mol×K	744.62	Joback Method
cpg	365.43	J/mol×K	780.85	Joback Method

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C1129415&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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