

# Diethylmalonic acid, heptadecyl 2,2,3,3,3-pentafluoropropyl ester

**Inchi:** InChI=1S/C27H47F5O4/c1-4-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-35-23(33)25(5)  
**InchiKey:** SWNXFVFJSIVWEF-UHFFFAOYSA-N  
**Formula:** C27H47F5O4  
**SMILES:** CCCCCCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCC(F)(F)C(F)(F)F  
**Mol. weight [g/mol]:** 530.65

## Physical Properties

Property code	Value	Unit	Source
gf	-1256.91	kJ/mol	Joback Method
hf	-2097.01	kJ/mol	Joback Method
hfus	64.42	kJ/mol	Joback Method
hvap	86.03	kJ/mol	Joback Method
log10ws	-9.58		Crippen Method
logp	8.948		Crippen Method
mvol	415.020	ml/mol	McGowan Method
pc	662.21	kPa	Joback Method
rinpol	2499.00		NIST Webbook
rinpol	2499.00		NIST Webbook
tb	956.40	K	Joback Method
tc	1186.65	K	Joback Method
tf	548.58	K	Joback Method
vc	1.653	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1442.75	J/mol×K	956.40	Joback Method
cpg	1464.27	J/mol×K	994.77	Joback Method
cpg	1484.28	J/mol×K	1033.15	Joback Method
cpg	1502.93	J/mol×K	1071.52	Joback Method
cpg	1520.37	J/mol×K	1109.90	Joback Method
cpg	1536.73	J/mol×K	1148.27	Joback Method
cpg	1552.17	J/mol×K	1186.65	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370854&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370854&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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