

# Atropic acid

<b>Other names:</b>	2-Phenylacrylic acid 2-Propenoic acid, 2-phenyl- Acrylic acid, 2-phenyl- Benzeneacetic acid, «alpha»-methylene- Benzeneacetic acid, Â«alphaÂ»-methylene- NSC 20987 Propenoic acid, 2-phenyl- «alpha»-Phenylacrylic acid «alpha»-Toluic acid, «alpha»-methylene- Â«alphaÂ»-Phenylacrylic acid Â«alphaÂ»-Toluic acid, Â«alphaÂ»-methylene-
<b>Inchi:</b>	InChI=1S/C9H8O2/c1-7(9(10)11)8-5-3-2-4-6-8/h2-6H,1H2,(H,10,11)
<b>InchiKey:</b>	ONPJWQSDZCGSQM-UHFFFAOYSA-N
<b>Formula:</b>	C9H8O2
<b>SMILES:</b>	<chem>C=C(C(=O)O)c1ccccc1</chem>
<b>Mol. weight [g/mol]:</b>	148.16
<b>CAS:</b>	492-38-6

## Physical Properties

Property code	Value	Unit	Source
gf	-49.14	kJ/mol	Joback Method
hf	-141.73	kJ/mol	Joback Method
hfus	16.20	kJ/mol	Joback Method
hvap	60.74	kJ/mol	Joback Method
log10ws	-2.06		Aqueous Solubility Prediction Method
logp	1.784		Crippen Method
mcvol	117.050	ml/mol	McGowan Method
pc	4167.71	kPa	Joback Method
tb	574.61	K	Joback Method
tc	785.54	K	Joback Method
tf	379.65	K	Aqueous Solubility Prediction Method
vc	0.439	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	261.34	J/mol×K	574.61	Joback Method
cpg	271.27	J/mol×K	609.77	Joback Method
cpg	280.51	J/mol×K	644.92	Joback Method
cpg	289.10	J/mol×K	680.08	Joback Method
cpg	297.10	J/mol×K	715.23	Joback Method
cpg	304.52	J/mol×K	750.39	Joback Method
cpg	311.40	J/mol×K	785.54	Joback Method

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C492386&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/25-535-0/Atropic-acid.pdf>

Generated by Cheméo on 2024-04-30 09:04:20.890382969 +0000 UTC m=+16757109.810960281.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.