

# Fumaric acid, isobutyl pent-4-enyl ester

<b>Inchi:</b>	InChI=1S/C13H20O4/c1-4-5-6-9-16-12(14)7-8-13(15)17-10-11(2)3/h4,7-8,11H,1,5-6,9-10
<b>InchiKey:</b>	HHXQOEXPOISSEC-BQYQJAHWSA-N
<b>Formula:</b>	C13H20O4
<b>SMILES:</b>	C=CCCCOC(=O)C=CC(=O)OCC(C)C
<b>Mol. weight [g/mol]:</b>	240.30

## Physical Properties

Property code	Value	Unit	Source
gf	-243.64	kJ/mol	Joback Method
hf	-563.88	kJ/mol	Joback Method
hfus	30.40	kJ/mol	Joback Method
hvap	61.74	kJ/mol	Joback Method
log10ws	-2.45		Crippen Method
logp	2.251		Crippen Method
mvol	200.310	ml/mol	McGowan Method
pc	1937.24	kPa	Joback Method
rinpol	1627.00		NIST Webbook
rinpol	1627.00		NIST Webbook
tb	649.82	K	Joback Method
tc	837.70	K	Joback Method
tf	358.75	K	Joback Method
vc	0.766	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	524.80	J/molxK	649.82	Joback Method
cpg	539.15	J/molxK	681.13	Joback Method
cpg	552.77	J/molxK	712.45	Joback Method
cpg	565.68	J/molxK	743.76	Joback Method
cpg	577.90	J/molxK	775.07	Joback Method
cpg	589.43	J/molxK	806.39	Joback Method
cpg	600.30	J/molxK	837.70	Joback Method
dvisc	0.0018071	Paxs	358.75	Joback Method

dvisc	0.0008647	Paxs	407.26	Joback Method
dvisc	0.0004841	Paxs	455.77	Joback Method
dvisc	0.0003030	Paxs	504.28	Joback Method
dvisc	0.0002059	Paxs	552.80	Joback Method
dvisc	0.0001489	Paxs	601.31	Joback Method
dvisc	0.0001130	Paxs	649.82	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348846&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348846&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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