

# Hydroperoxide, hexyl

<b>Other names:</b>	Hexyl hydroperoxide n-C6H13OOH 1-Hydroperoxyhexane
<b>Inchi:</b>	InChI=1S/C6H14O2/c1-2-3-4-5-6-8-7/h7H,2-6H2,1H3
<b>InchiKey:</b>	RZICEOJUAFHYFO-UHFFFAOYSA-N
<b>Formula:</b>	C6H14O2
<b>SMILES:</b>	CCCCCOO
<b>Mol. weight [g/mol]:</b>	118.17
<b>CAS:</b>	4312-76-9

## Physical Properties

Property code	Value	Unit	Source
chl	-4062.00 ± 5.00	kJ/mol	NIST Webbook
gf	-344.27	kJ/mol	Joback Method
hf	-517.44	kJ/mol	Joback Method
hfus	21.44	kJ/mol	Joback Method
hvap	46.11	kJ/mol	Joback Method
ie	9.47 ± 0.03	eV	NIST Webbook
ie	9.85	eV	NIST Webbook
log10ws	-1.74		Crippen Method
logp	2.056		Crippen Method
mcvol	107.140	ml/mol	McGowan Method
pc	3195.54	kPa	Joback Method
tb	482.19	K	Joback Method
tc	671.20	K	Joback Method
tf	317.98	K	Joback Method
vc	0.396	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	225.87	J/mol×K	482.19	Joback Method
cpg	271.15	J/mol×K	639.70	Joback Method
cpg	262.98	J/mol×K	608.20	Joback Method

cpg	254.38	J/molxK	576.70	Joback Method
cpg	245.34	J/molxK	545.19	Joback Method
cpg	235.84	J/molxK	513.69	Joback Method
cpg	278.92	J/molxK	671.20	Joback Method
dvisc	0.0002997	Paxs	482.19	Joback Method
dvisc	0.0003680	Paxs	454.82	Joback Method
dvisc	0.0004638	Paxs	427.45	Joback Method
dvisc	0.0006034	Paxs	400.09	Joback Method
dvisc	0.0008160	Paxs	372.72	Joback Method
dvisc	0.0011576	Paxs	345.35	Joback Method
dvisc	0.0017439	Paxs	317.98	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4312769&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4312769&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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