

# 2H-Pyran, 2-(bromomethyl)tetrahydro-

<b>Other names:</b>	2-(Bromomethyl)tetrahydro-2H-pyran Pyran, 2-(bromomethyl)tetrahydro-
<b>Inchi:</b>	InChI=1S/C6H11BrO/c7-5-6-3-1-2-4-8-6/h6H,1-5H2
<b>InchiKey:</b>	MHNWCBOXPOLLIB-UHFFFAOYSA-N
<b>Formula:</b>	C6H11BrO
<b>SMILES:</b>	BrCC1CCCCO1
<b>Mol. weight [g/mol]:</b>	179.06
<b>CAS:</b>	34723-82-5

## Physical Properties

Property code	Value	Unit	Source
gf	-47.71	kJ/mol	Joback Method
hf	-218.52	kJ/mol	Joback Method
hfus	16.39	kJ/mol	Joback Method
hvap	40.32	kJ/mol	Joback Method
log10ws	-1.86		Crippen Method
logp	1.950		Crippen Method
mcvol	107.910	ml/mol	McGowan Method
pc	4277.45	kPa	Joback Method
tb	426.20	K	NIST Webbook
tc	674.65	K	Joback Method
tf	251.13	K	Joback Method
vc	0.388	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	205.43	J/molxK	449.34	Joback Method
cpg	268.70	J/molxK	637.10	Joback Method
cpg	257.63	J/molxK	599.55	Joback Method
cpg	245.80	J/molxK	562.00	Joback Method
cpg	233.17	J/molxK	524.44	Joback Method
cpg	219.73	J/molxK	486.89	Joback Method
cpg	279.04	J/molxK	674.65	Joback Method

dvisc	0.0004172	Paxs	449.34	Joback Method
dvisc	0.0005449	Paxs	416.31	Joback Method
dvisc	0.0007450	Paxs	383.27	Joback Method
dvisc	0.0010807	Paxs	350.24	Joback Method
dvisc	0.0016939	Paxs	317.20	Joback Method
dvisc	0.0029474	Paxs	284.17	Joback Method
dvisc	0.0059330	Paxs	251.13	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C34723825&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C34723825&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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