

# **(CH<sub>3</sub>)<sub>2</sub>N-C(CH<sub>3</sub>)=N(n-C<sub>3</sub>H<sub>7</sub>)**

Inchi:	InChI=1S/C7H16N2/c1-5-6-8-7(2)9(3)4/h5-6H2,1-4H3
InchiKey:	VCEKZPKWRURQCS-UHFFFAOYSA-N
Formula:	C7H16N2
SMILES:	CCCN=C(C)N(C)C
Mol. weight [g/mol]:	128.22
CAS:	94793-20-1

## **Physical Properties**

Property code	Value	Unit	Source
affp	1030.30	kJ/mol	NIST Webbook
basg	997.90	kJ/mol	NIST Webbook
hf	-47.85	kJ/mol	Joback Method
hvap	36.61	kJ/mol	Joback Method
log10ws	-0.98		Crippen Method
logp	1.376		Crippen Method
mcvol	125.150	ml/mol	McGowan Method
pc	2520.12	kPa	Joback Method
tb	448.56	K	Joback Method
tc	636.15	K	Joback Method

## **Sources**

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C94793201&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C94793201&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## **Legend**

**affp:** Proton affinity

<b>basg:</b>	Gas basicity
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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