

Propanoic acid, 2-hydroxy-, pentyl ester

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| Other names: | amyl lactate lactic acid, pentyl ester n-Amyl lactate pentyl 2-hydroxypropanoate pentyl lactate |
| Inchi: | InChI=1S/C8H16O3/c1-3-4-5-6-11-8(10)7(2)9/h7,9H,3-6H2,1-2H3 |
| InchiKey: | GXOHBWLPQHLYPF-UHFFFAOYSA-N |
| Formula: | C8H16O3 |
| SMILES: | CCCCCOC(=O)C(C)O |
| Mol. weight [g/mol]: | 160.21 |
| CAS: | 6382-06-5 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------------|---------|----------------|
| gf | -356.70 | kJ/mol | Joback Method |
| hf | -610.76 | kJ/mol | Joback Method |
| hfus | 19.83 | kJ/mol | Joback Method |
| hvap | 58.85 | kJ/mol | Joback Method |
| log10ws | -1.41 | | Crippen Method |
| logp | 1.101 | | Crippen Method |
| mcvol | 136.890 | ml/mol | McGowan Method |
| pc | 2953.69 | kPa | Joback Method |
| rinpol | 1097.00 | | NIST Webbook |
| rinpol | 1097.00 | | NIST Webbook |
| rinpol | 1075.00 | | NIST Webbook |
| rinpol | 1075.00 | | NIST Webbook |
| ripol | 1610.00 | | NIST Webbook |
| ripol | 1610.00 | | NIST Webbook |
| tb | 550.47 | K | Joback Method |
| tc | 722.09 | K | Joback Method |
| tf | 251.00 ± 2.00 | K | NIST Webbook |
| vc | 0.520 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---|
| cpg | 393.29 | J/molxK | 722.09 | Joback Method |
| cpg | 344.69 | J/molxK | 579.07 | Joback Method |
| cpg | 355.26 | J/molxK | 607.68 | Joback Method |
| cpg | 365.40 | J/molxK | 636.28 | Joback Method |
| cpg | 375.11 | J/molxK | 664.88 | Joback Method |
| cpg | 384.41 | J/molxK | 693.49 | Joback Method |
| cpg | 333.69 | J/molxK | 550.47 | Joback Method |
| dvisc | 0.0001680 | Paxs | 508.38 | Joback Method |
| dvisc | 0.0002959 | Paxs | 466.28 | Joback Method |
| dvisc | 0.0005834 | Paxs | 424.19 | Joback Method |
| dvisc | 0.0013358 | Paxs | 382.09 | Joback Method |
| dvisc | 0.0037547 | Paxs | 340.00 | Joback Method |
| dvisc | 0.0001040 | Paxs | 550.47 | Joback Method |
| dvisc | 0.0141338 | Paxs | 297.90 | Joback Method |
| hvapt | 73.90 | kJ/mol | 378.50 | NIST Webbook |
| pvap | 0.07 | kPa | 315.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.09 | kPa | 319.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.11 | kPa | 321.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |

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|------|------|-----|--------|---|
| pvap | 0.06 | kPa | 313.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.14 | kPa | 325.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.17 | kPa | 327.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.19 | kPa | 329.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.22 | kPa | 331.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |

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|------|------|-----|--------|---|
| pvap | 0.25 | kPa | 333.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.28 | kPa | 335.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.31 | kPa | 337.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.35 | kPa | 339.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.42 | kPa | 341.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |

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|------|------|-----|--------|---|
| pvap | 0.47 | kPa | 343.20 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.05 | kPa | 311.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.04 | kPa | 309.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.04 | kPa | 307.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.03 | kPa | 305.30 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |

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|------|------|-----|--------|---|
| pvap | 0.03 | kPa | 303.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.08 | kPa | 317.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |
| pvap | 0.13 | kPa | 323.40 | Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods |

Sources

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| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Renewable platform chemicals: Evaluation of thermochemical data of alkyl lactates with complementary experimental and computational methods: | https://www.doi.org/10.1016/j.jct.2018.07.029 |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C6382065&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |

Legend

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| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |

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| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| hvapt: | Enthalpy of vaporization at a given temperature |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| pvap: | Vapor pressure |
| rinpol: | Non-polar retention indices |
| ripol: | Polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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