

# 2-Ethylbutyric acid, phenyl ester

|                      |                                                                           |
|----------------------|---------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C12H16O2/c1-3-10(4-2)12(13)14-11-8-6-5-7-9-11/h5-10H,3-4H2,1-2H3 |
| InchiKey:            | QTQSMZUPHLTSKC-UHFFFAOYSA-N                                               |
| Formula:             | C12H16O2                                                                  |
| SMILES:              | CCC(CC)C(=O)Oc1ccccc1                                                     |
| Mol. weight [g/mol]: | 192.25                                                                    |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -73.79  | kJ/mol  | Joback Method  |
| hf            | -304.56 | kJ/mol  | Joback Method  |
| hfus          | 20.14   | kJ/mol  | Joback Method  |
| hvap          | 53.35   | kJ/mol  | Joback Method  |
| log10ws       | -3.22   |         | Crippen Method |
| logp          | 3.028   |         | Crippen Method |
| mcvol         | 163.620 | ml/mol  | McGowan Method |
| pc            | 2537.93 | kPa     | Joback Method  |
| rinpol        | 1398.00 |         | NIST Webbook   |
| tb            | 576.49  | K       | Joback Method  |
| tc            | 786.80  | K       | Joback Method  |
| tf            | 308.58  | K       | Joback Method  |
| vc            | 0.618   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 394.41    | J/mol×K | 576.49          | Joback Method |
| cpg           | 409.96    | J/mol×K | 611.54          | Joback Method |
| cpg           | 424.63    | J/mol×K | 646.59          | Joback Method |
| cpg           | 438.42    | J/mol×K | 681.64          | Joback Method |
| cpg           | 451.37    | J/mol×K | 716.69          | Joback Method |
| cpg           | 463.49    | J/mol×K | 751.74          | Joback Method |
| cpg           | 474.82    | J/mol×K | 786.80          | Joback Method |
| dvisc         | 0.0029260 | Pa×s    | 308.58          | Joback Method |
| dvisc         | 0.0013546 | Pa×s    | 353.23          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0007455 | Pa×s | 397.88 | Joback Method |
| dvisc | 0.0004628 | Pa×s | 442.54 | Joback Method |
| dvisc | 0.0003136 | Pa×s | 487.19 | Joback Method |
| dvisc | 0.0002268 | Pa×s | 531.84 | Joback Method |
| dvisc | 0.0001725 | Pa×s | 576.49 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357751&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357751&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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