

# Heptafluorobutyric anhydride

<b>Other names:</b>	Heptafluorobutanoic anhydride Perfluorobutyric anhydride Heptafluoro-n-butyric anhydride Butanoic acid, heptafluoro-, anhydride
<b>Inchi:</b>	InChI=1S/C8F14O3/c9-3(10,5(13,14)7(17,18)19)1(23)25-2(24)4(11,12)6(15,16)8(20,21)2
<b>InchiKey:</b>	UFFSXJKVKBQEHK-UHFFFAOYSA-N
<b>Formula:</b>	C8F14O3
<b>SMILES:</b>	O=C(OC(=O)C(F)(F)C(F)(F)C(F)(F)F)C(F)(F)C(F)(F)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	410.06
<b>CAS:</b>	336-59-4

## Physical Properties

Property code	Value	Unit	Source
gf	-3056.66	kJ/mol	Joback Method
hf	-3363.87	kJ/mol	Joback Method
hfus	19.50	kJ/mol	Joback Method
hvap	30.09	kJ/mol	Joback Method
log10ws	-4.39		Crippen Method
logp	3.722		Crippen Method
mcvol	157.370	ml/mol	McGowan Method
pc	1746.28	kPa	Joback Method
tb	382.20	K	NIST Webbook
tb	379.50 ± 0.50	K	NIST Webbook
tb	382.00 ± 1.00	K	NIST Webbook
tc	619.92	K	Joback Method
tf	324.79	K	Joback Method
vc	0.700	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	433.28	J/mol×K	483.00	Joback Method
cpg	443.32	J/mol×K	505.82	Joback Method
cpg	452.61	J/mol×K	528.64	Joback Method

cpg	461.17	J/mol×K	551.46	Joback Method
cpg	469.04	J/mol×K	574.28	Joback Method
cpg	476.25	J/mol×K	597.10	Joback Method
cpg	482.86	J/mol×K	619.92	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C336594&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C336594&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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