

Phenol, 2,6-dimethyl-4-nitro-

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| Other names: | 2,6-Dimethyl-4-nitrophenol 2,6-Xylenol, 4-nitro- 4-Nitro-2,6-dimethylphenol 4-Nitro-2,6-xylenol |
| Inchi: | InChI=1S/C8H9NO3/c1-5-3-7(9(11)12)4-6(2)8(5)10/h3-4,10H,1-2H3 |
| InchiKey: | FNORUNUDZNWQFF-UHFFFAOYSA-N |
| Formula: | C8H9NO3 |
| SMILES: | <chem>Cc1cc([N+](=O)[O-])cc(C)c1O</chem> |
| Mol. weight [g/mol]: | 167.16 |
| CAS: | 2423-71-4 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -9.44 | kJ/mol | Joback Method |
| hf | -182.93 | kJ/mol | Joback Method |
| hfus | 26.88 | kJ/mol | Joback Method |
| hvap | 66.61 | kJ/mol | Joback Method |
| log10ws | -2.62 | | Crippen Method |
| logp | 1.917 | | Crippen Method |
| mcvol | 123.110 | ml/mol | McGowan Method |
| pc | 4305.56 | kPa | Joback Method |
| rinpol | 1680.00 | | NIST Webbook |
| rinpol | 1680.00 | | NIST Webbook |
| tb | 651.54 | K | Joback Method |
| tc | 906.38 | K | Joback Method |
| tf | 486.71 | K | Joback Method |
| vc | 0.423 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 306.80 | J/mol×K | 651.54 | Joback Method |
| cpg | 317.13 | J/mol×K | 694.01 | Joback Method |
| cpg | 326.73 | J/mol×K | 736.49 | Joback Method |

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|-----|--------|---------|--------|---------------|
| cpg | 335.71 | J/mol×K | 778.96 | Joback Method |
| cpg | 344.15 | J/mol×K | 821.43 | Joback Method |
| cpg | 352.17 | J/mol×K | 863.91 | Joback Method |
| cpg | 359.86 | J/mol×K | 906.38 | Joback Method |

Correlations

| Information | Value |
|-----------------------------|-------------------------------|
| Property code | pvap |
| Equation | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A | 8.59786e+00 |
| Coeff. B | -1.88985e+03 |
| Coeff. C | -8.39840e+01 |
| Temperature range (K), min. | 311.40 |
| Temperature range (K), max. | 659.04 |

Sources

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|---|---|
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C2423714&Units=SI |
| The Yaws Handbook of Vapor Pressure: | https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |

| | |
|----------------|----------------------------------|
| pc: | Critical Pressure |
| pvap: | Vapor pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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