

# Perfluoro-1-heptene

|                             |  |
|-----------------------------|--|
| <b>Other names:</b>         | 1-Heptene, 1,1,2,3,3,4,4,5,5,6,6,7,7,7-tetradecafluoro-Hept-1-ene, perfluoro Perfluoro heptene-1 Tetradecafluoro-1-heptene perfluorohept-1-ene |
| <b>Inchi:</b>               | InChI=1S/C7F14/c8-1(2(9)10)3(11,12)4(13,14)5(15,16)6(17,18)7(19,20)21  |
| <b>InchiKey:</b>            | CDAVUOSPHHTNBU-UHFFFAOYSA-N  |
| <b>Formula:</b>             | C7F14  |
| <b>SMILES:</b>              | FC(F)=C(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)F   |
| <b>Mol. weight [g/mol]:</b> | 350.05   |
| <b>CAS:</b>                 | 355-63-5   |

## Physical Properties

| Property code | Value         | Unit                 | Source         |
|---------------|---------------|----------------------|----------------|
| gf            | -2641.96      | kJ/mol               | Joback Method  |
| hf            | -2879.46      | kJ/mol               | Joback Method  |
| hfus          | 17.52         | kJ/mol               | Joback Method  |
| hvap          | 13.38         | kJ/mol               | Joback Method  |
| ie            | 10.48 ± 0.02  | eV                   | NIST Webbook   |
| log10ws       | -5.57         |                      | Crippen Method |
| logp          | 5.168         |                      | Crippen Method |
| mcvol         | 129.970       | ml/mol               | McGowan Method |
| pc            | 1790.91       | kPa                  | Joback Method  |
| rinpol        | 312.00        |                      | NIST Webbook   |
| tb            | 354.00 ± 3.00 | K                    | NIST Webbook   |
| tb            | 354.00 ± 1.00 | K                    | NIST Webbook   |
| tc            | 453.23        | K                    | Joback Method  |
| tf            | 156.01        | K                    | Joback Method  |
| vc            | 0.607         | m <sup>3</sup> /kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 292.90 | J/mol×K | 337.11          | Joback Method |

|     |        |         |        |               |
|-----|--------|---------|--------|---------------|
| cpg | 304.92 | J/mol×K | 356.46 | Joback Method |
| cpg | 316.25 | J/mol×K | 375.82 | Joback Method |
| cpg | 326.90 | J/mol×K | 395.17 | Joback Method |
| cpg | 336.90 | J/mol×K | 414.52 | Joback Method |
| cpg | 346.27 | J/mol×K | 433.87 | Joback Method |
| cpg | 355.04 | J/mol×K | 453.23 | Joback Method |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.50264e+01                   |
| Coeff. B                    | -3.27224e+03                  |
| Coeff. C                    | -3.96040e+01                  |
| Temperature range (K), min. | 261.62                        |
| Temperature range (K), max. | 376.43                        |

## Sources

|   |   |
|---|---|
| <b>Crippen Method:</b>                      | <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>   |
| <b>Crippen Method:</b>                      | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>   |
| <b>Joback Method:</b>                       | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>   |
| <b>McGowan Method:</b>                      | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>   |
| <b>NIST Webbook:</b>                        | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C355635&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C355635&amp;Units=SI</a>   |
| <b>The Yaws Handbook of Vapor Pressure:</b> | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |

## Legend

|              |   |
|--------------|---|
| <b>cpg:</b>  | Ideal gas heat capacity                         |
| <b>gf:</b>   | Standard Gibbs free energy of formation         |
| <b>hf:</b>   | Enthalpy of formation at standard conditions    |
| <b>hfus:</b> | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b> | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>   | Ionization energy                               |

|                 |                                     |
|-----------------|-------------------------------------|
| <b>log10ws:</b> | Log10 of Water solubility in mol/l  |
| <b>logp:</b>    | Octanol/Water partition coefficient |
| <b>mcvol:</b>   | McGowan's characteristic volume     |
| <b>pc:</b>      | Critical Pressure                   |
| <b>pvap:</b>    | Vapor pressure                      |
| <b>rinpola:</b> | Non-polar retention indices         |
| <b>tb:</b>      | Normal Boiling Point Temperature    |
| <b>tc:</b>      | Critical Temperature                |
| <b>tf:</b>      | Normal melting (fusion) point       |
| <b>vc:</b>      | Critical Volume                     |

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