

# Diethyl bis(hydroxymethyl)malonate

<b>Other names:</b>	Propanedioic acid, bis(hydroxymethyl)-, diethyl ester
<b>Inchi:</b>	InChI=1S/C9H16O6/c1-3-14-7(12)9(5-10,6-11)8(13)15-4-2/h10-11H,3-6H2,1-2H3
<b>InchiKey:</b>	WIOHBOKEUIHYIC-UHFFFAOYSA-N
<b>Formula:</b>	C9H16O6
<b>SMILES:</b>	CCOC(=O)C(CO)(CO)C(=O)OCC
<b>Mol. weight [g/mol]:</b>	220.22
<b>CAS:</b>	20605-01-0

## Physical Properties

Property code	Value	Unit	Source
gf	-713.74	kJ/mol	Joback Method
hf	-1031.90	kJ/mol	Joback Method
hfus	25.40	kJ/mol	Joback Method
hvap	86.00	kJ/mol	Joback Method
log10ws	0.40		Crippen Method
logp	-0.916		Crippen Method
mcvol	164.290	ml/mol	McGowan Method
pc	3131.49	kPa	Joback Method
tb	739.03	K	Joback Method
tc	919.46	K	Joback Method
tf	459.57	K	Joback Method
vc	0.615	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	474.01	J/molxK	739.03	Joback Method
cpg	515.28	J/molxK	889.39	Joback Method
cpg	508.07	J/molxK	859.32	Joback Method
cpg	500.35	J/molxK	829.25	Joback Method
cpg	492.11	J/molxK	799.17	Joback Method
cpg	483.33	J/molxK	769.10	Joback Method
cpg	521.97	J/molxK	919.46	Joback Method
dvisc	0.0000077	Paxs	739.03	Joback Method

dvisc	0.0000127	Paxs	692.45	Joback Method
dvisc	0.0000228	Paxs	645.88	Joback Method
dvisc	0.0000445	Paxs	599.30	Joback Method
dvisc	0.0000975	Paxs	552.72	Joback Method
dvisc	0.0002467	Paxs	506.15	Joback Method
dvisc	0.0007531	Paxs	459.57	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20605010&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20605010&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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