

# 2-Ethylbutyric acid, 4-benzyloxyphenyl ester

<b>Inchi:</b>	InChI=1S/C19H22O3/c1-3-16(4-2)19(20)22-18-12-10-17(11-13-18)21-14-15-8-6-5-7-9-15
<b>InchiKey:</b>	ZIDVZFPAIRHFJN-UHFFFAOYSA-N
<b>Formula:</b>	C19H22O3
<b>SMILES:</b>	CCC(CC)C(=O)Oc1ccc(OCc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	298.38

## Physical Properties

Property code	Value	Unit	Source
gf	-17.07	kJ/mol	Joback Method
hf	-356.20	kJ/mol	Joback Method
hfus	33.11	kJ/mol	Joback Method
hvap	74.28	kJ/mol	Joback Method
log10ws	-5.45		Crippen Method
logp	4.607		Crippen Method
mcvol	244.360	ml/mol	McGowan Method
pc	1798.51	kPa	Joback Method
rinpol	2420.00		NIST Webbook
tb	790.73	K	Joback Method
tc	1014.28	K	Joback Method
tf	448.64	K	Joback Method
vc	0.919	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	710.10	J/molxK	790.73	Joback Method
cpg	726.50	J/molxK	827.99	Joback Method
cpg	741.60	J/molxK	865.25	Joback Method
cpg	755.44	J/molxK	902.51	Joback Method
cpg	768.08	J/molxK	939.76	Joback Method
cpg	779.53	J/molxK	977.02	Joback Method
cpg	789.84	J/molxK	1014.28	Joback Method
dvisc	0.0007740	Paxs	448.64	Joback Method
dvisc	0.0003983	Paxs	505.66	Joback Method

dvisc	0.0002345	Paxs	562.67	Joback Method
dvisc	0.0001522	Paxs	619.69	Joback Method
dvisc	0.0001063	Paxs	676.70	Joback Method
dvisc	0.0000784	Paxs	733.72	Joback Method
dvisc	0.0000605	Paxs	790.73	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357757&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357757&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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